

The Eighth International Particle Accelerator Conference

Discovery of the Island of Stability for Super Heavy Elements

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Joint Institute for Nuclear Research

IPAC'17
May19, 2017. Copenhagen, Denmark

For more than 22 centuries

From Democritus (460 – 371 BC...)

to Dalton (1766 – 1844)

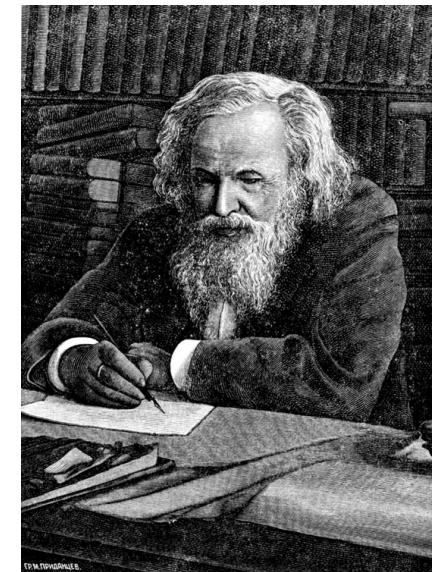
it was assumed that the objects around us are made up of tiny indivisible particles – atoms interacting with each other.

In 1808

36 elements were known only

63 elements were already known by that time

Ряды	Элементы								
	0	I	II	III	IV	V	VI	VII	
0	Натрий								
1	Короний	Водород H 1,008	—	—	—	—	—	—	
2	Гелий He 4,0	Литий Li 7,03	Бериллий Be 9,1	Бор B 11,0	Углерод C 12,0	Азот N 14,01	Кислород O 16,00	Фтор F 19,0	
3	Неон Ne 19,9	Натрий Na 23,05	Магний Mg 24,36	Алюминий Al 27,1	Кремний Si 28,2	Фосфор P 31,0	Сера S 32,06	Хлор Cl 35,45	
4	Аргон Ar 38	Калий K 39,15	Кальций Ca 40,1	Силиций Si 44,1	Титан Ti 48,1	Ванадий V 51,2	Хром Cr 52,1	Марганец Mn 55,1	Железо Fe 55,9
5		Медь Cu 63,6	Цинк Zn 65,4	Галлий Ga 70,0	Германий Ge 72,5	Мышьяк As 75	Селен Se 79,2	Бром Br 79,95	Кобальт Co 59
6	Краситон Kr 81,8	Рубидий Rb 85,5	Стронций Sr 87,6	Иттербий Y 89,0	Цирконий Zr 90,6	Ниобий Nb 94,0	Молибден Mo 96,0	—	Рутений Ru 101,7
7		Серебро Ag 107,93	Кадмий Cd 112,4	Индий In 115,0	Олово Sn 119,0	Сурьма Sb 120,2	Тантал Ta 127	Иод I 127	Родий Rh 103,0
8	Ксенон Xe 128	Цезий Cs 132,9	Барий Ba 137,4	Лантан La 138,9	Церий Ce 140,2	—	—	—	—
9		—	—	—	—	—	—	—	—
10	—	—	—	Иттербий Yb 173	—	Тантал Ta 183	Вольфрам W 184	—	Осиев Os 191
11									Иридий Ir 193
12	—	—	Радий Ra 225	—	Торий Th 232,5	—	Уран U 238,5		Платин Pt 194,3

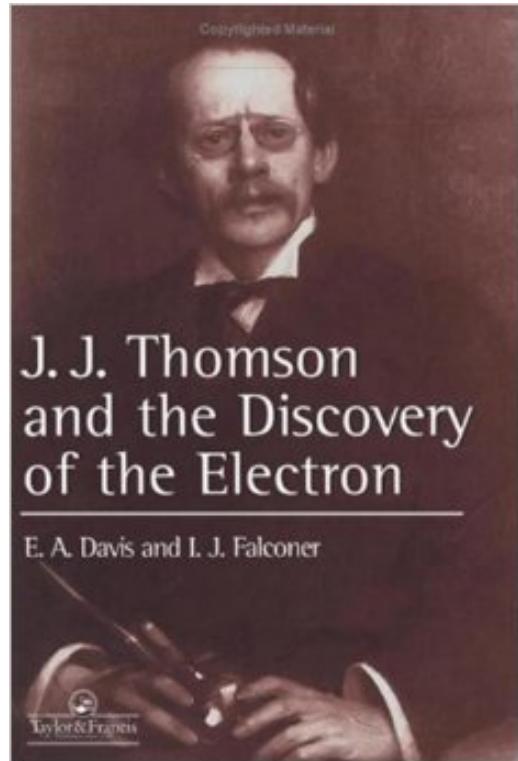
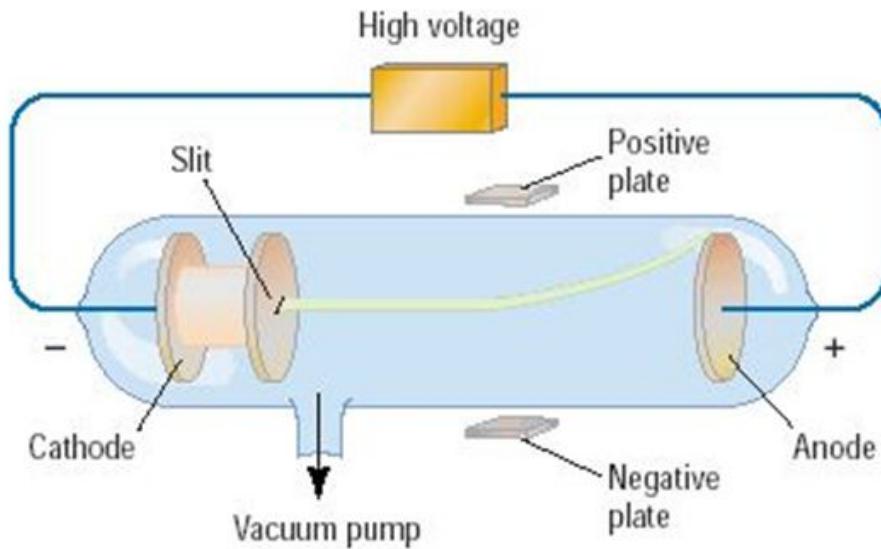


Dmitri Mendeleev
1869

Genuine D.I.Mendeleev's Table

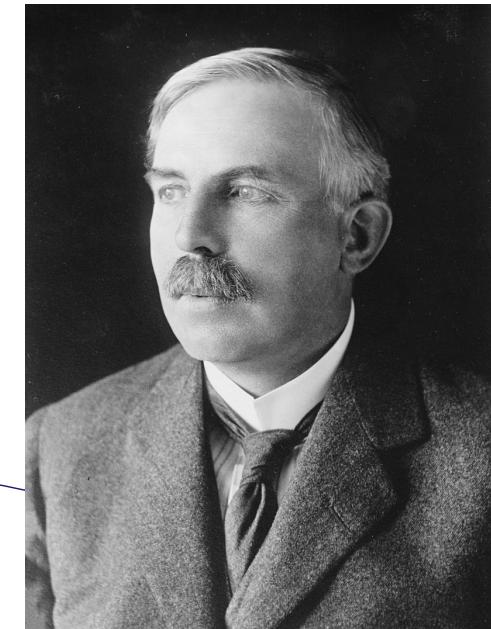
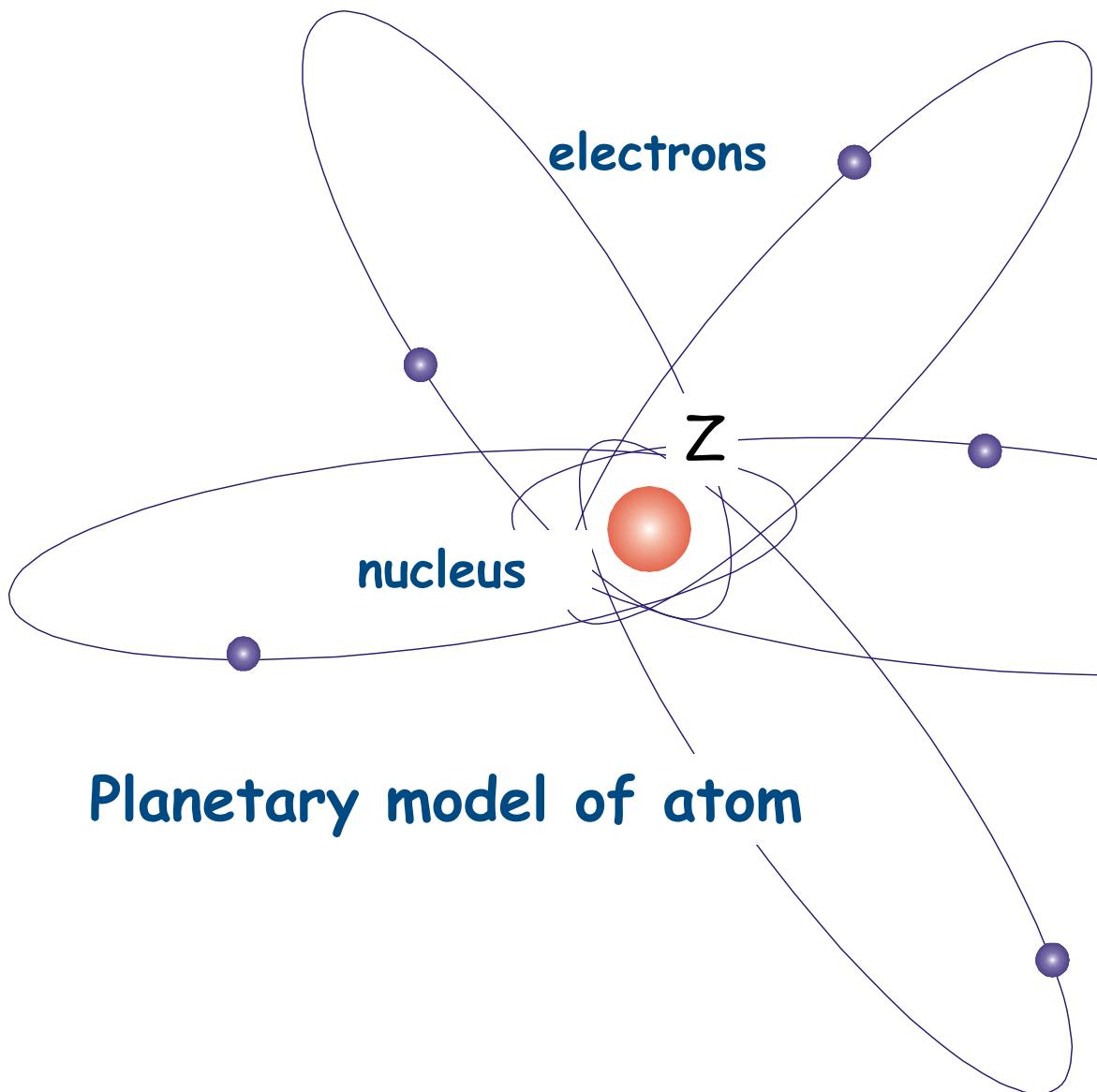
Discovery of the Electron

In 1897, J.J. Thomson used a cathode ray tube to deduce the presence of a negatively charged particle.



Cathode ray tubes pass electricity through a gas that is contained at a very low pressure.

86 elements were already known by that time



Ernest Rutherford
March 7, 1911

Planetary model of atom

In attempts to describe the properties of nuclear matter, George Gamow made a daring assumption that atomic nucleus may be similar to a drop of positively charged liquid.



George Gamow 1928

In fact, the density of nuclear liquid is 10^{15} times more than that of water

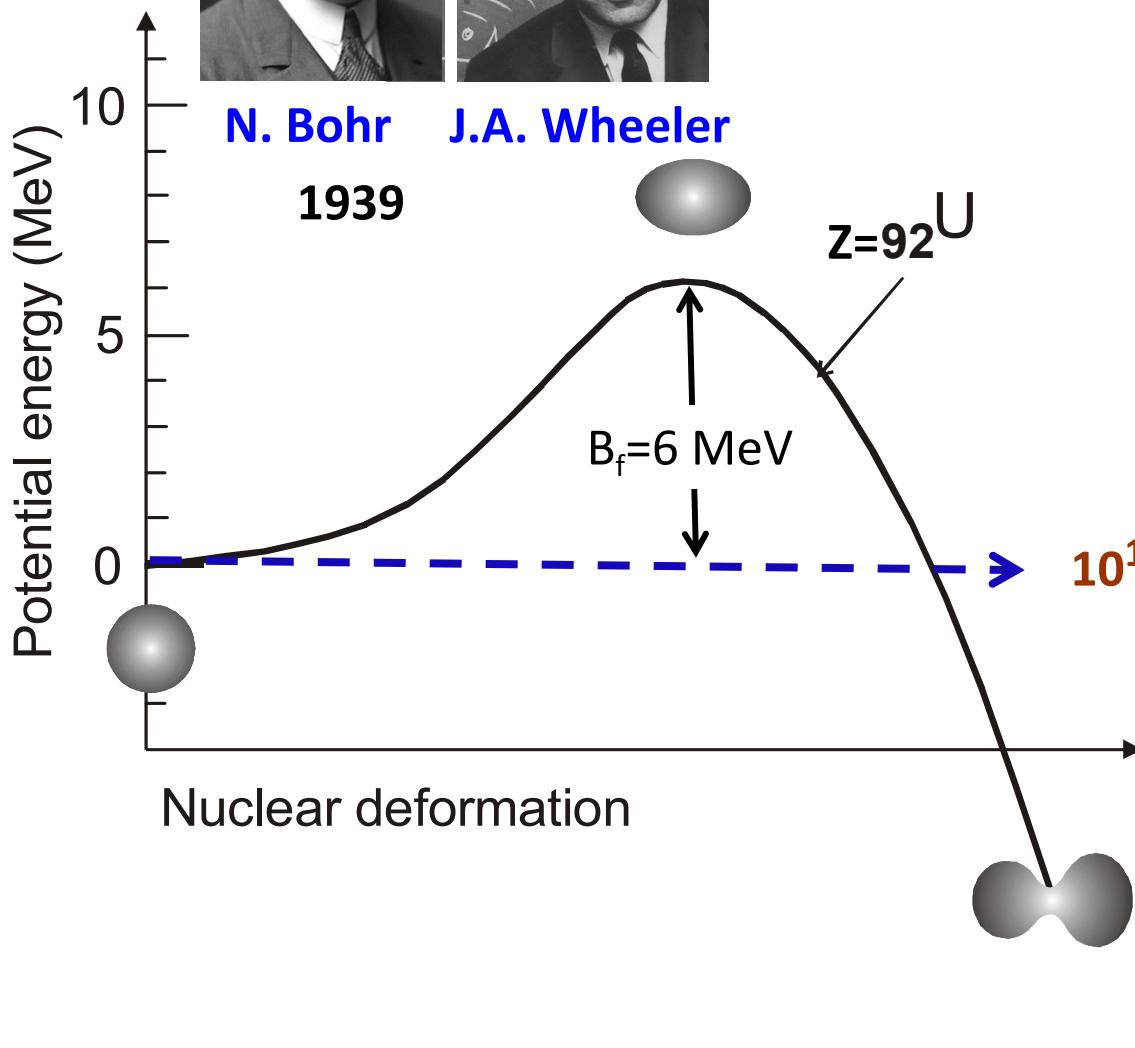
Charged Liquid Drop Model
of the atomic nucleus.

G. Gamow, 1928.

We will be discussing today the problem of synthesis and properties of extremely heavy nuclei to get answers to questions about:

- How big the nuclei can be?
- What of proton and neutron number they may have?
- Where is the mass limit of the nuclei?
- How many chemical elements can be and what are their properties?

Nuclear fission

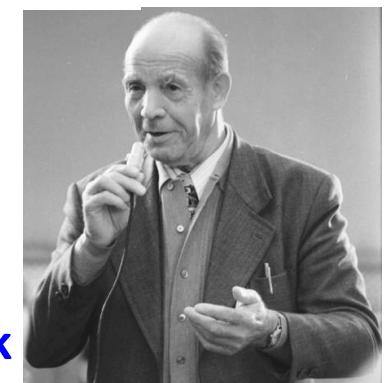


G.N. Flerov



1940

K.A. Petrzhak



Nuclear fission

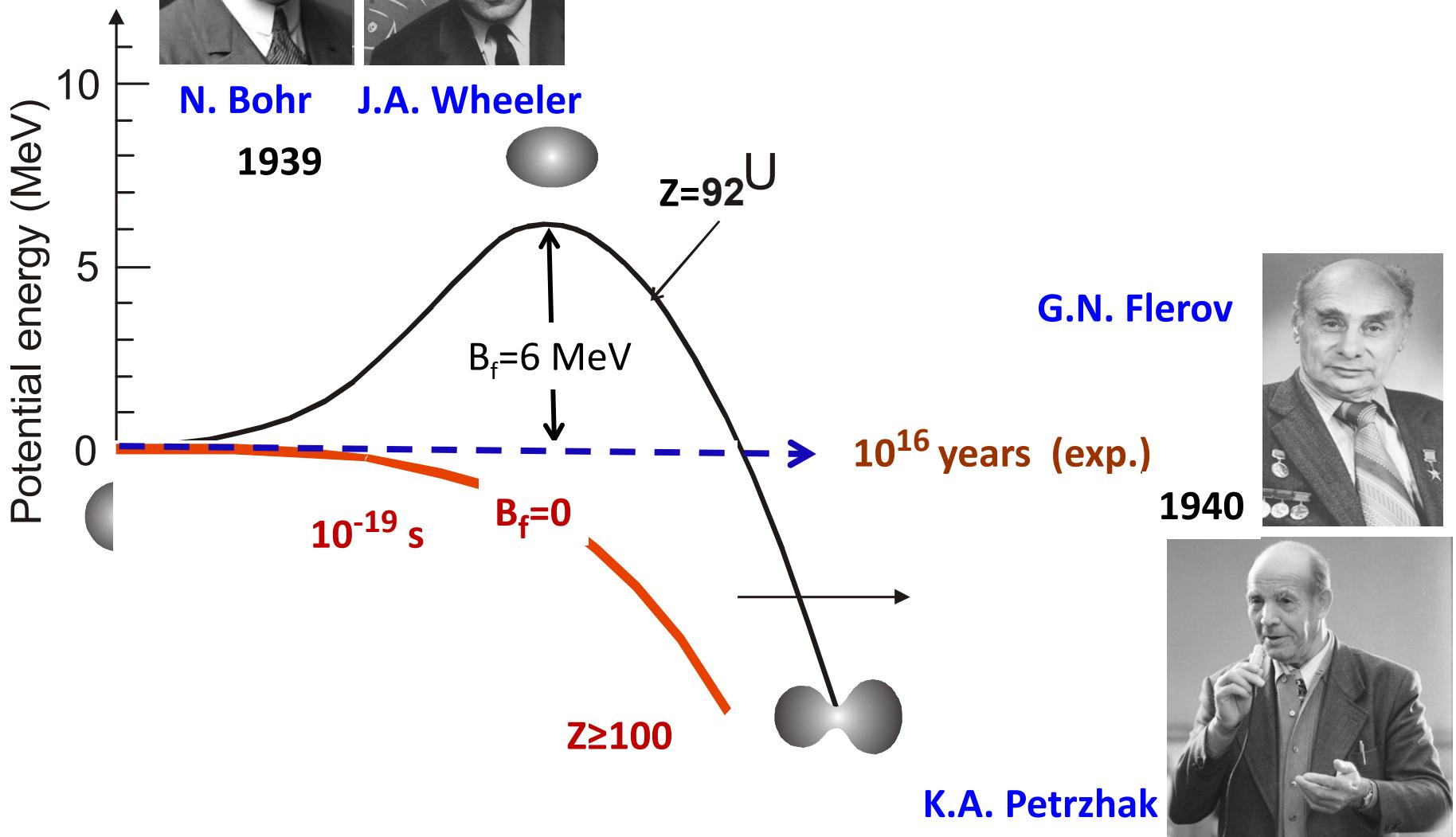
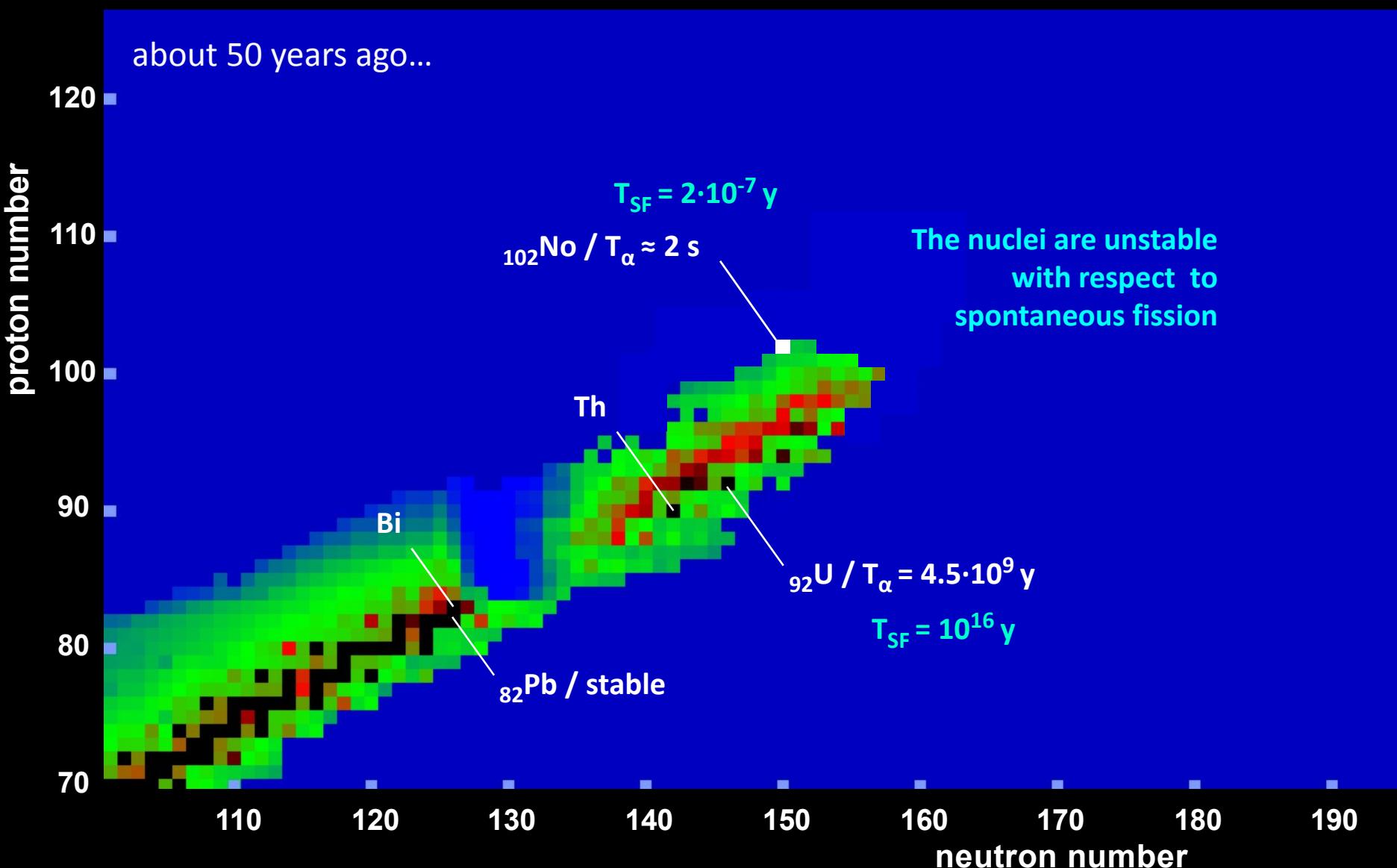
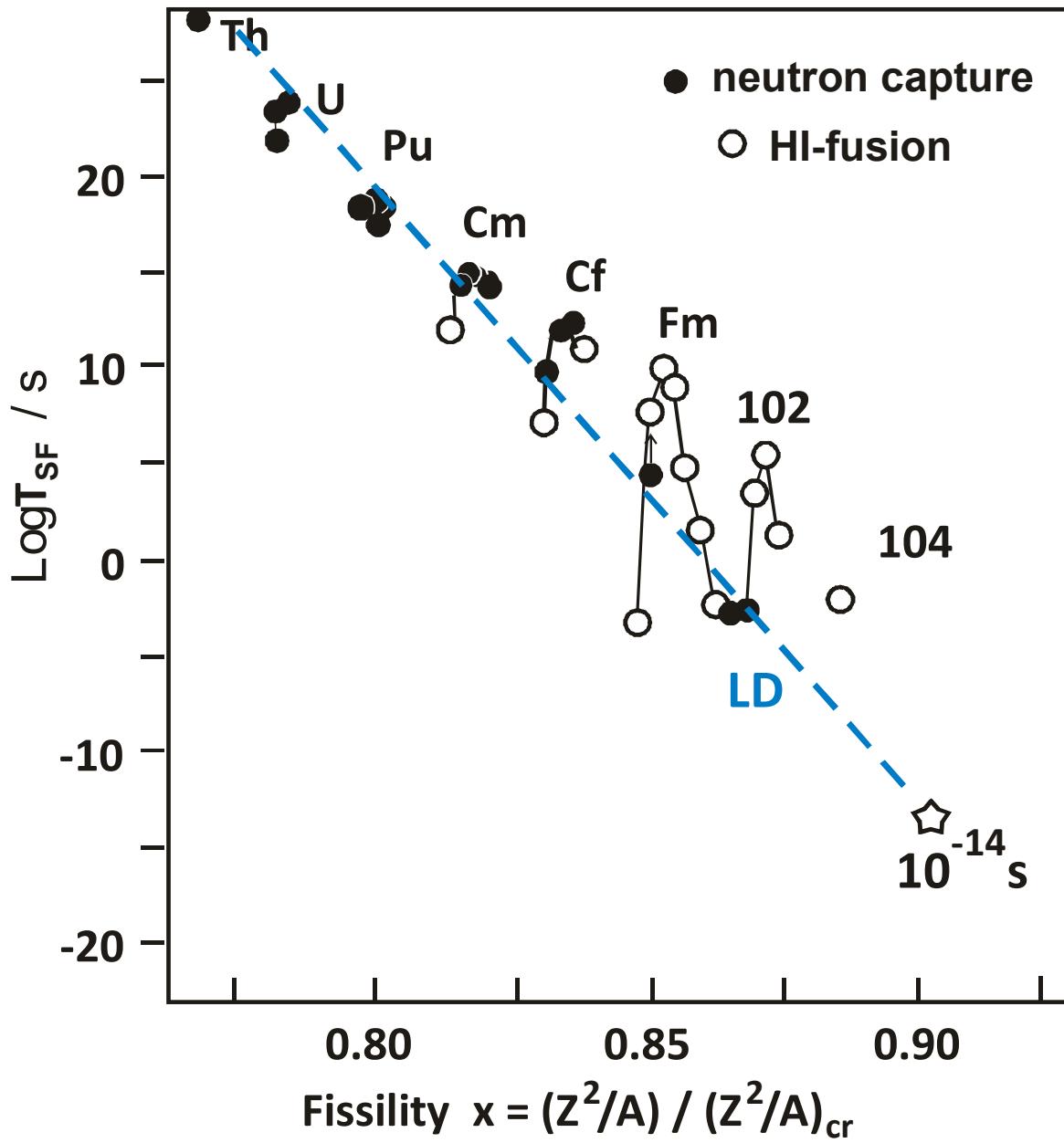


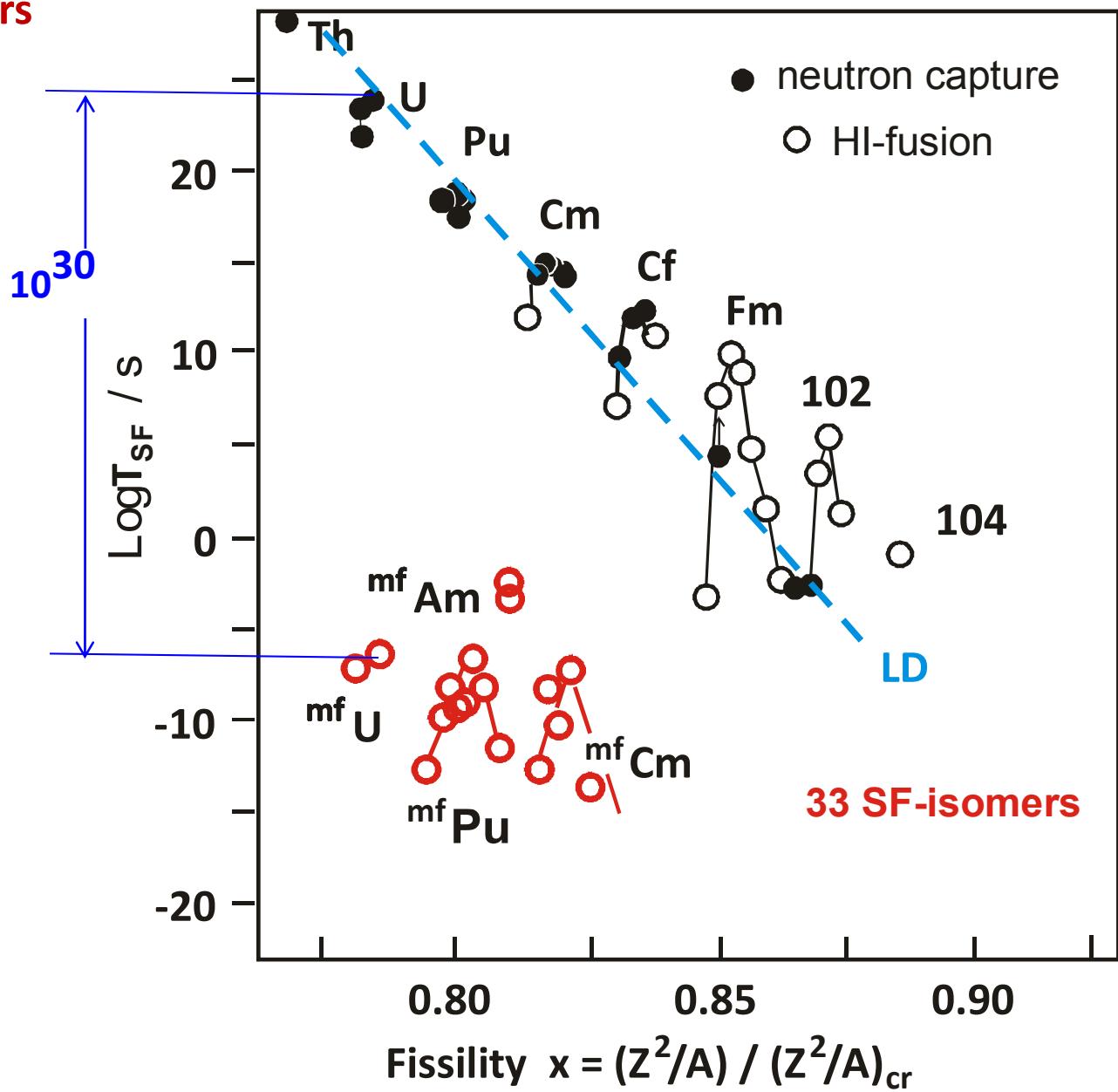
Chart of nuclides

Macroscopic theory (Liquid Drop Model)





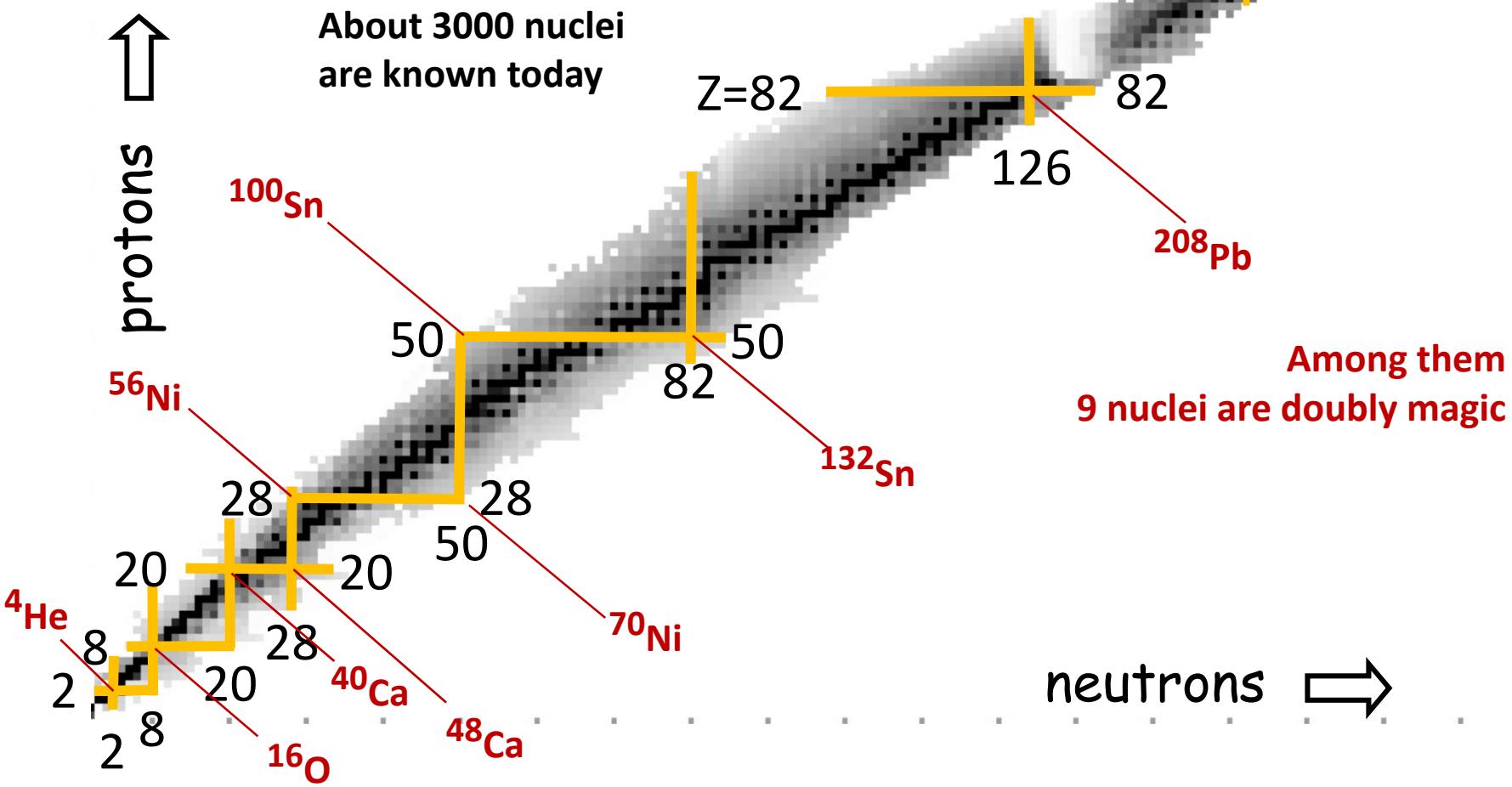
Discovery SF isomers JINR, Dubna, 1962



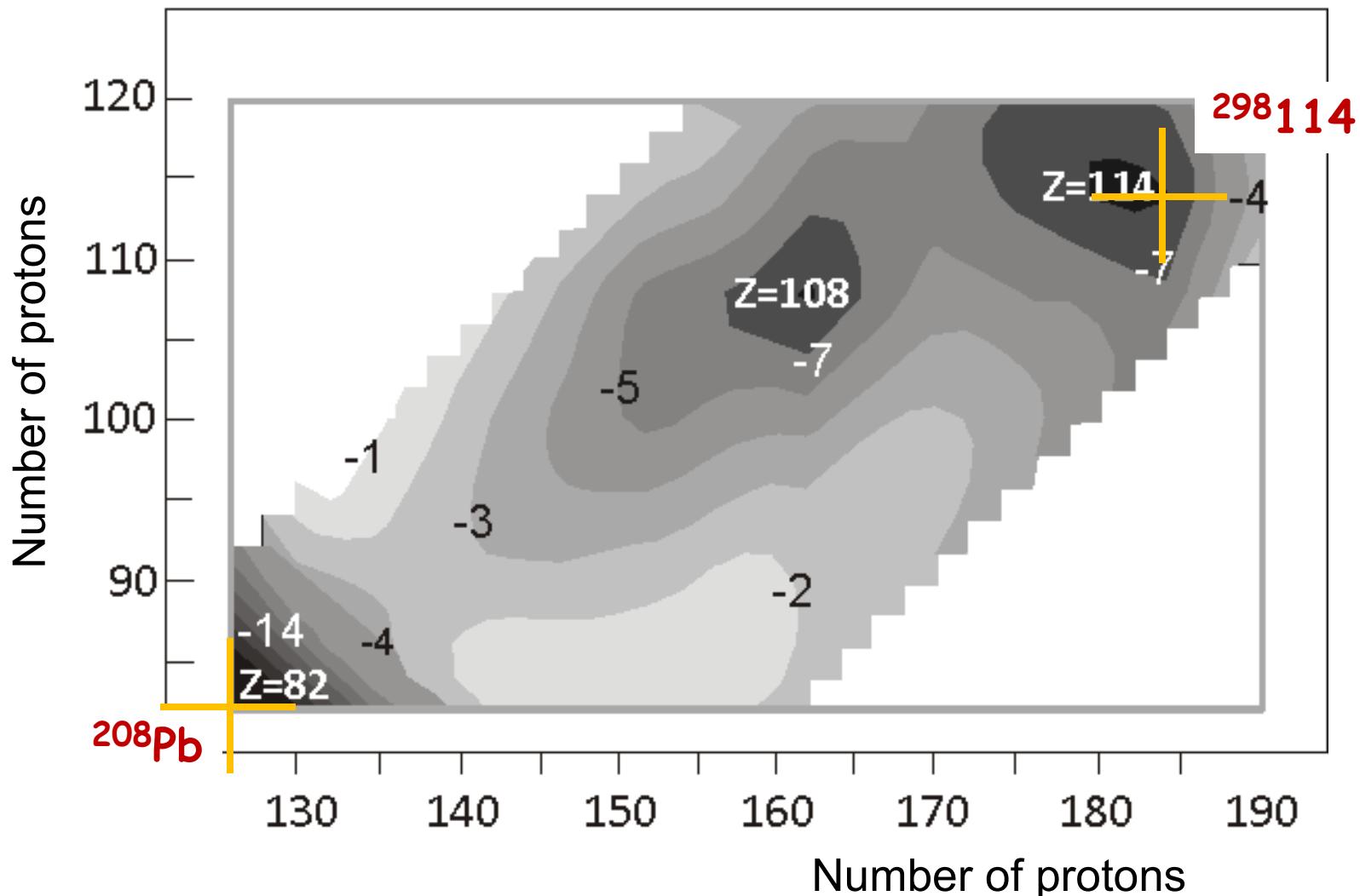
Shell and magic numbers

E. Wigner / M. Goeppert-Mayer / H. Jensen
Nobel Prize 1963

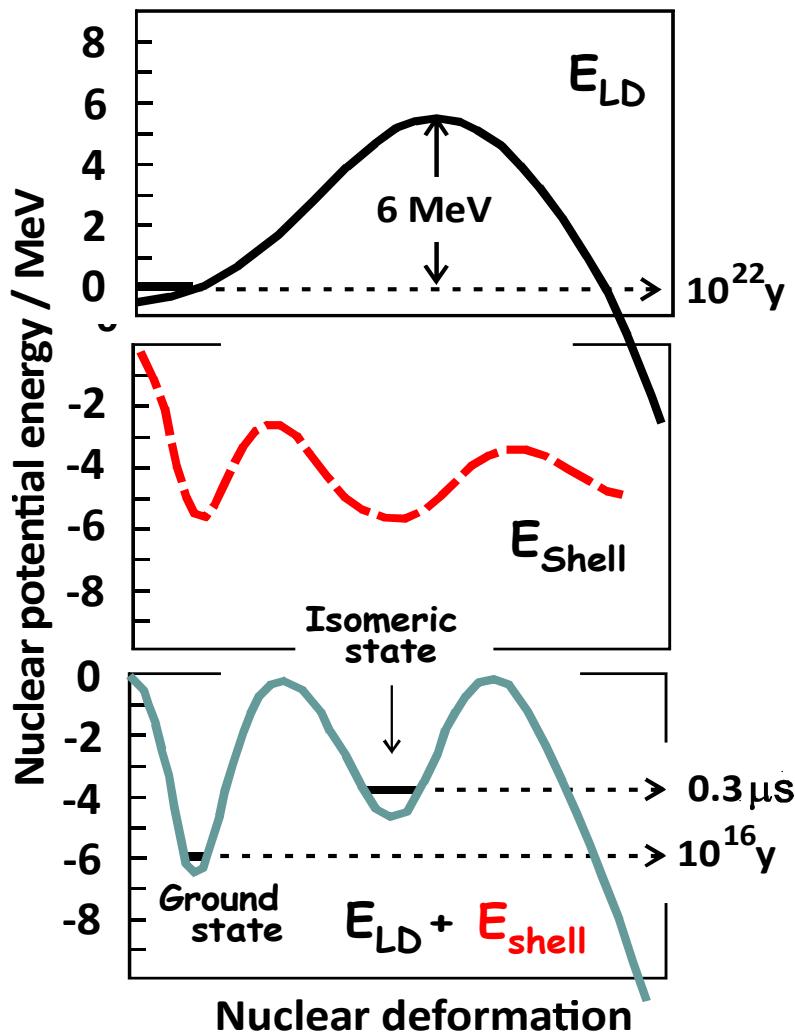
NEW SHELLS ?



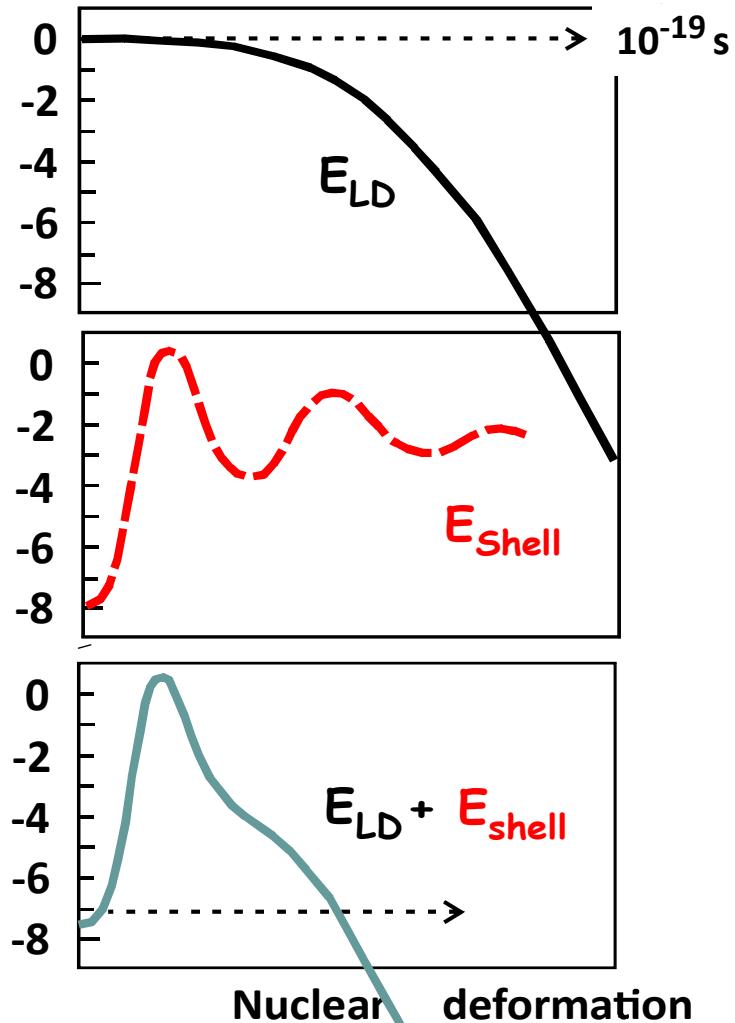
Map of the shell corrections (in MeV) to the liquid drop potential energy



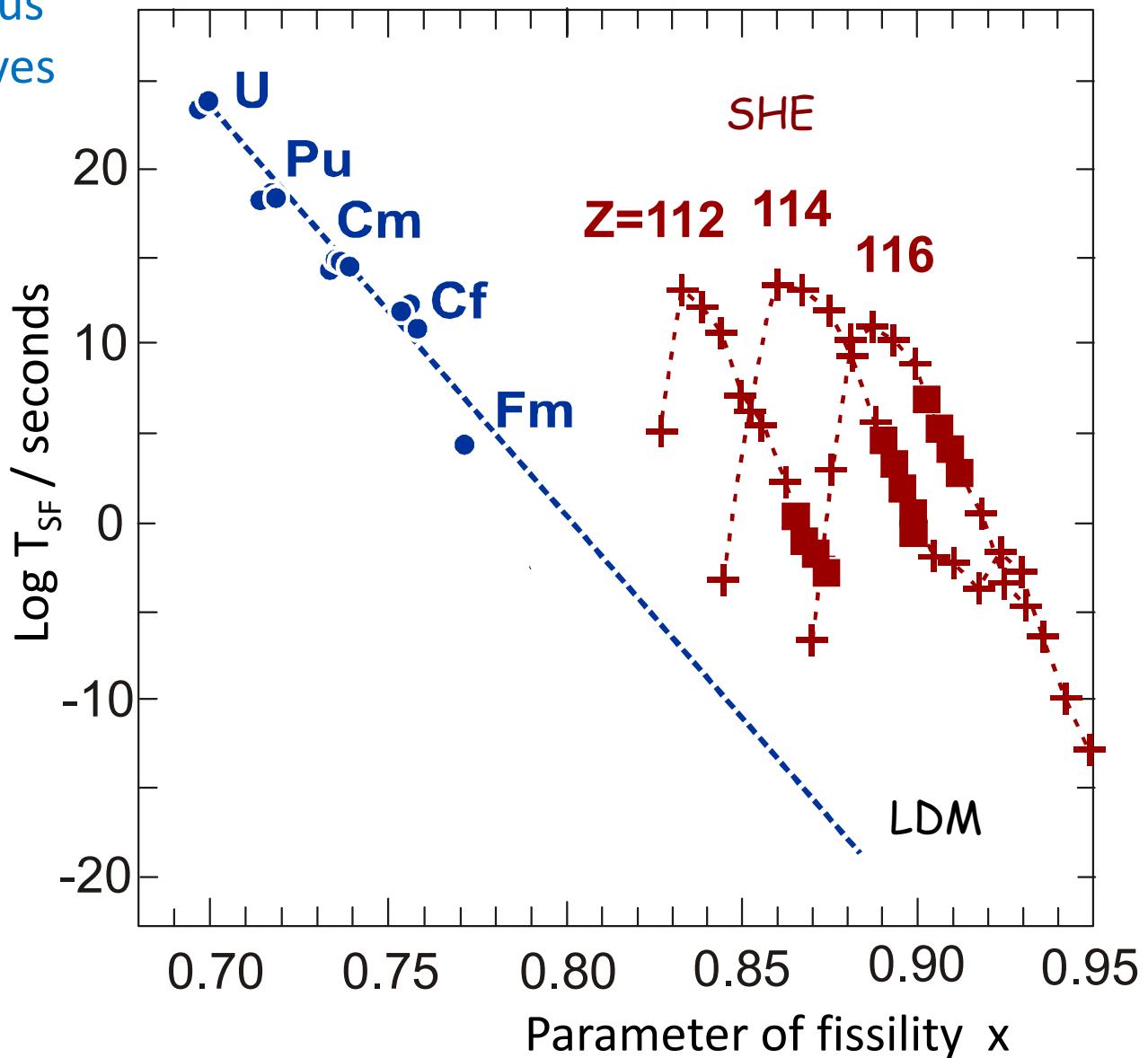
Uranium

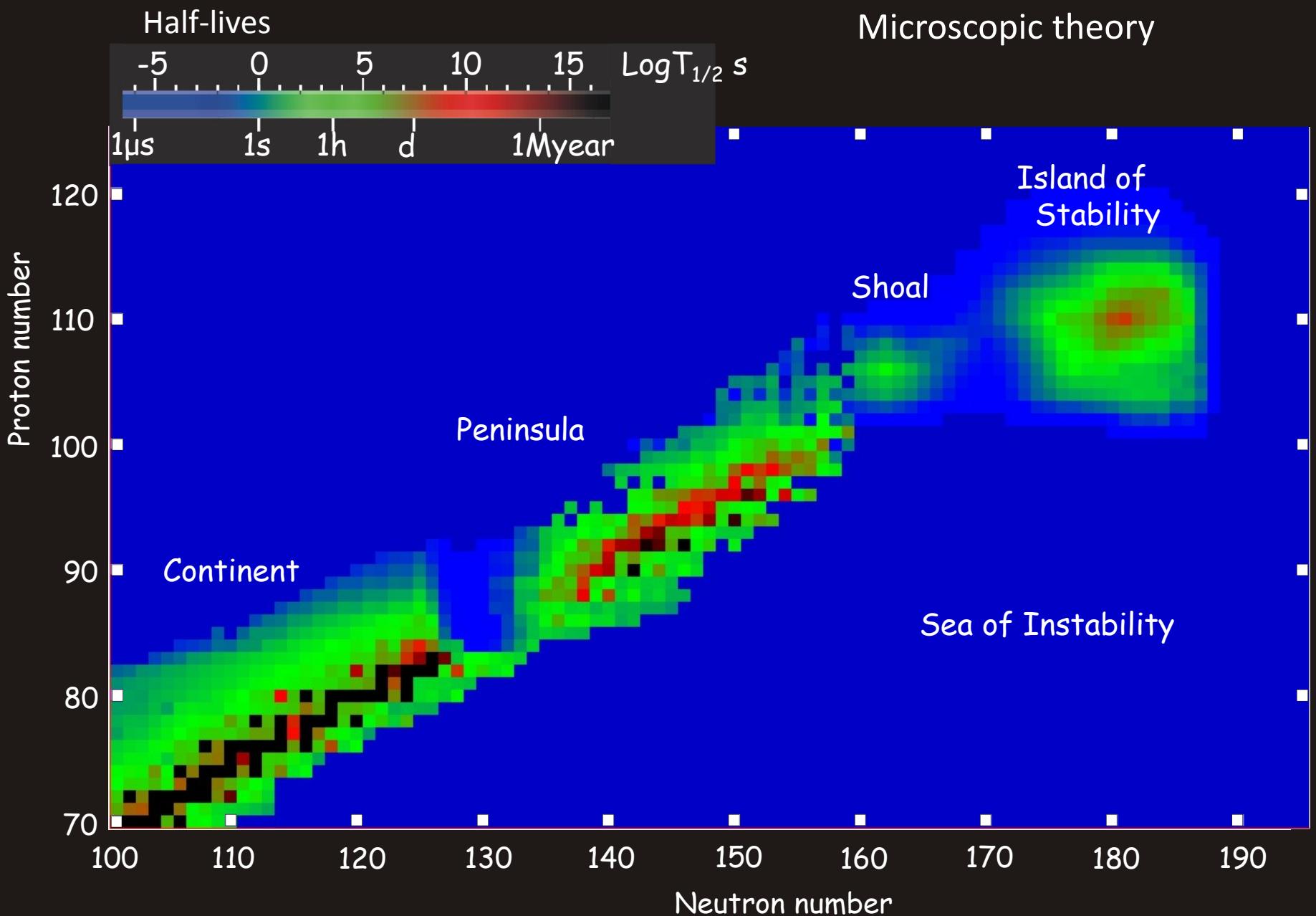


Super heavy nucleus



Partial Spontaneous Fission Half-Lives





Experimenters rushed to find this island

15-year long assault on the “Islands of Stability”

Los Alamos (USA)

1970-1985

Berkeley (USA)

Dubna (JINR)

Oak Ridge (USA)

Mainz (Germany)

Darmstadt (Germany)

Orsay (France)

Würenlingen (Switzerland)

Tokyo (Japan) some later

The task was:

To find the method of producing/detecting of superheavy elements

Search in nature: earth/lunar objects, cosmic rays,

artificial synthesis of superheavy elements

Use and develop a setup for synthesizing SHE

High-flux reactor, even nuclear explosion, powerful
accelerator of heavy ions,

To develop setups and detectors for separating and registering the
rarest events of formation and decay of superheavy nuclei

separator/detector, spectrometer, chemical methods, etc.

To have much patience

To work for many years, being ready to rearrange on the way and improve
the experimental technique

The task was:

To find the method of producing/detecting of elements
Search in nature: earth/lunar objects, artificial synthesis

Use and develop a setup for sv-

Higgs particles
SHE were not found
nuclear explosion, powerful accelerator of heavy ions,

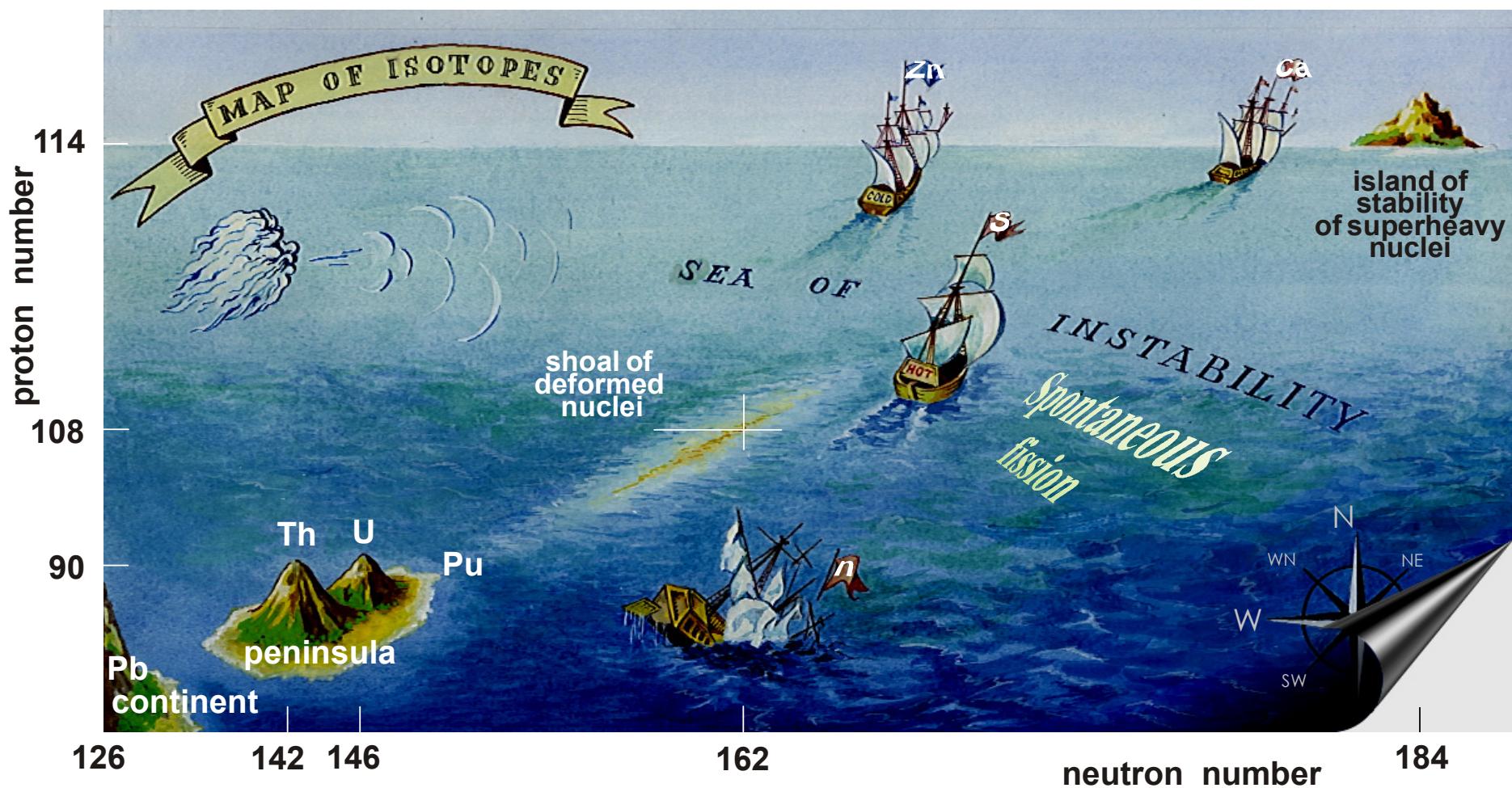
To develop setu-
rarest events

... separating and registering the decay of superheavy nuclei
, detector, spectrometer, chemical methods, etc.

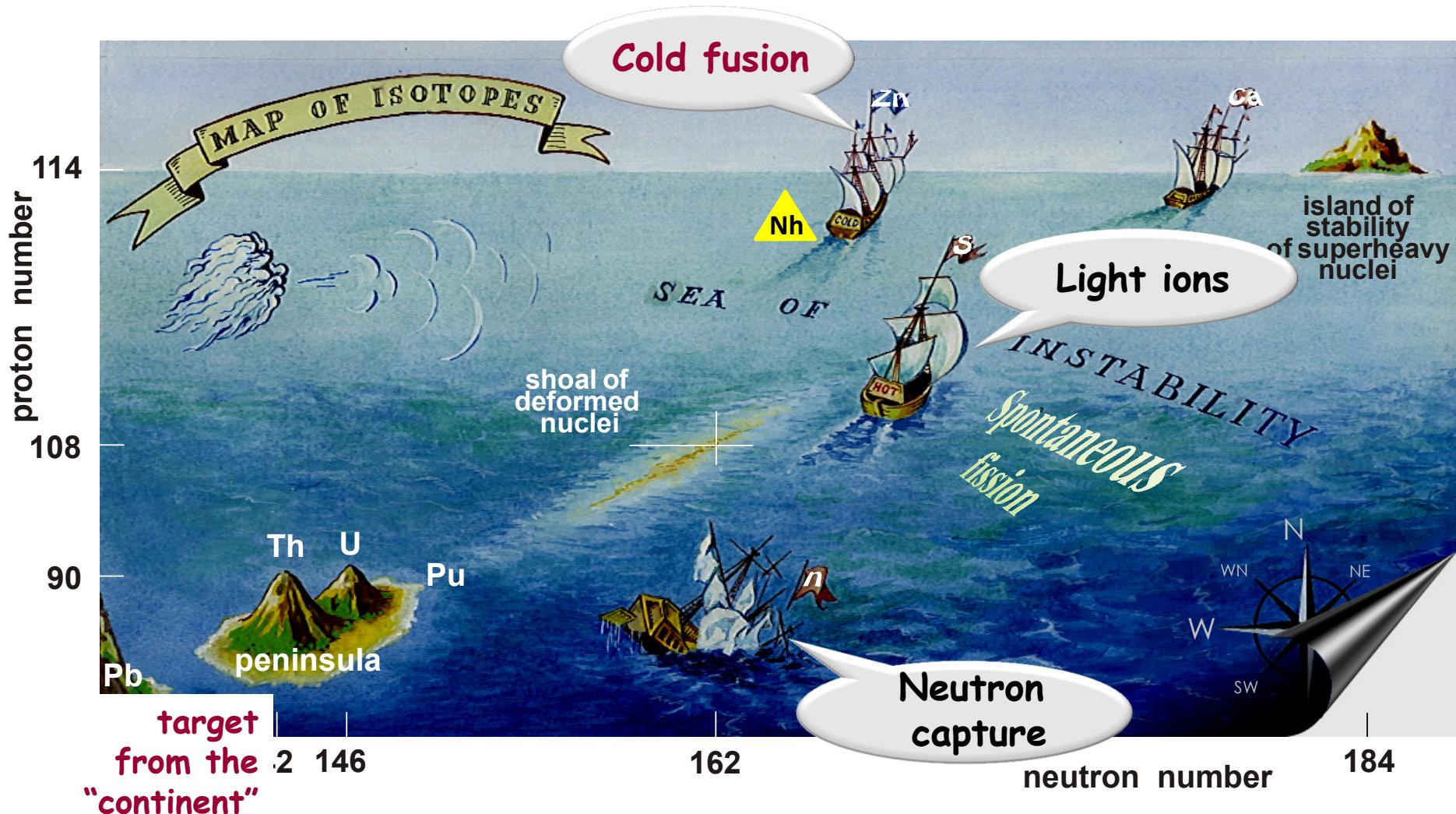
To have much time

To work for many years, being ready to rearrange on the way and improve the experimental technique

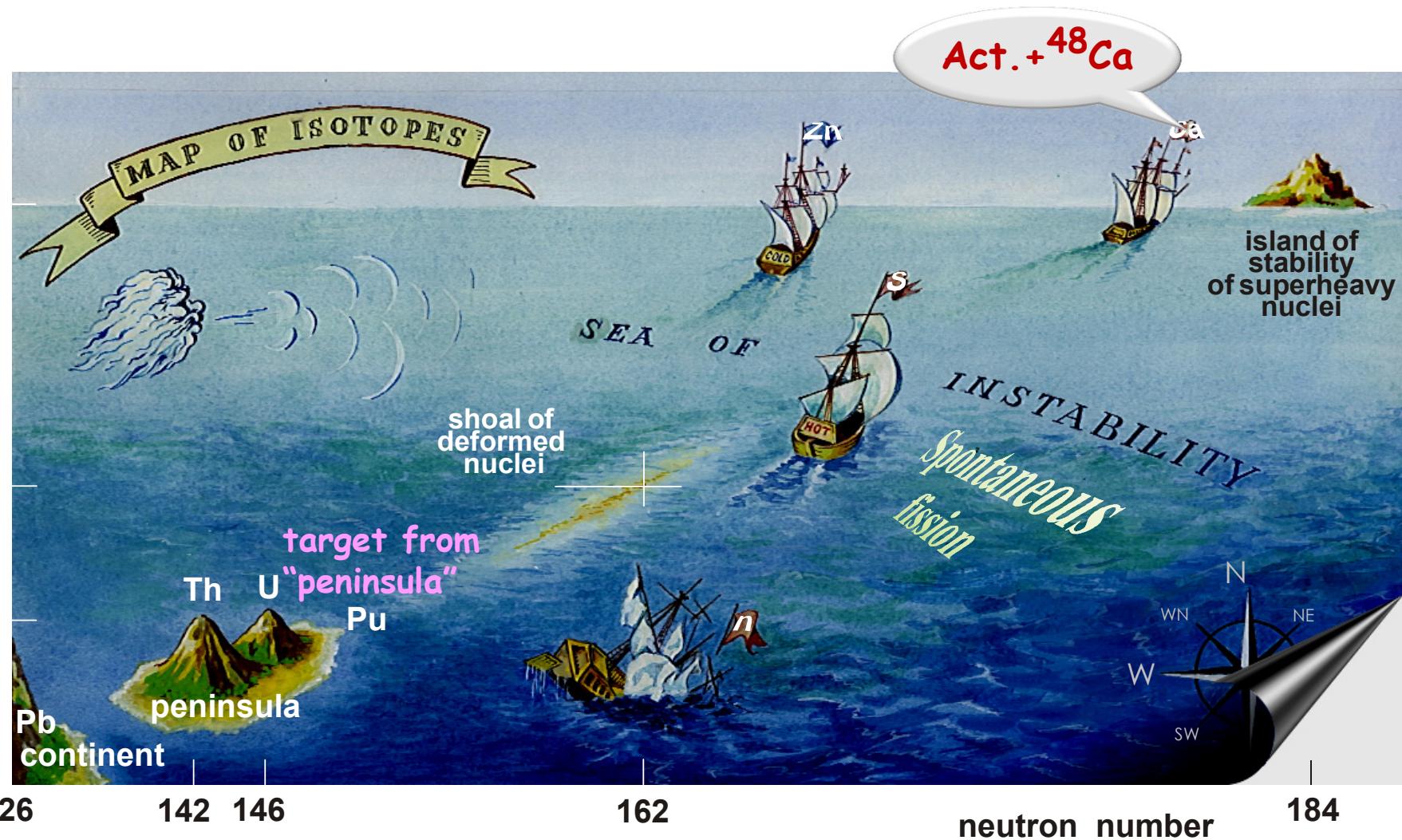
Reactions of synthesis



Reactions of synthesis



Our new approach greatly complicates the experiment



artificial element from
flux nuclear reactor

target

from
accelerator



to separator



projectiles

rare and very
expensive isotope of Ca

artificial element from
flux nuclear reactor

target



from
accelerator

projectiles

rare and very
expensive isotope of Ca

to separator

artificial element from
flux nuclear reactor

target



from
accelerator

projectiles

rare and very
expensive isotope of Ca

to separator

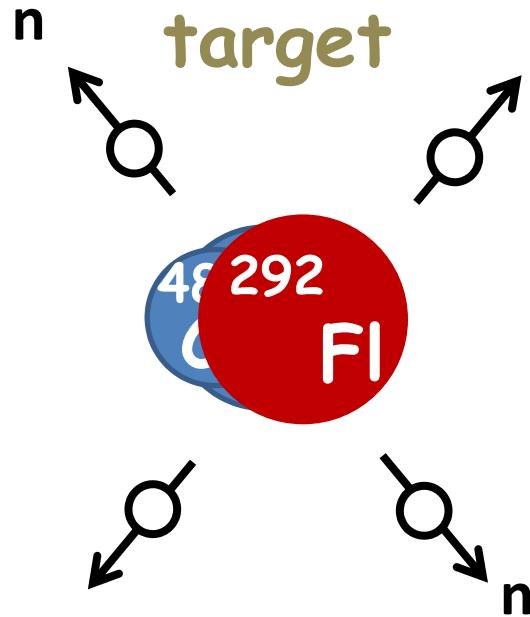
from
accelerator

projectiles

rare and very
expensive isotope of Ca

artificial element from
flux nuclear reactor

to separator



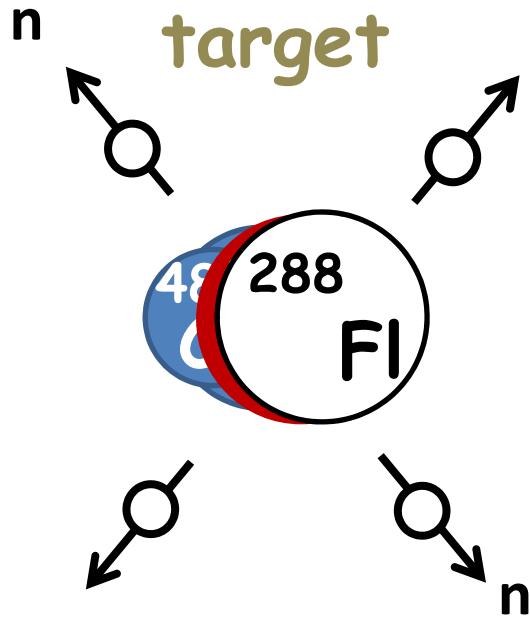
from
accelerator

projectiles

rare and very
expensive isotope of Ca

artificial element from
flux nuclear reactor

to separator

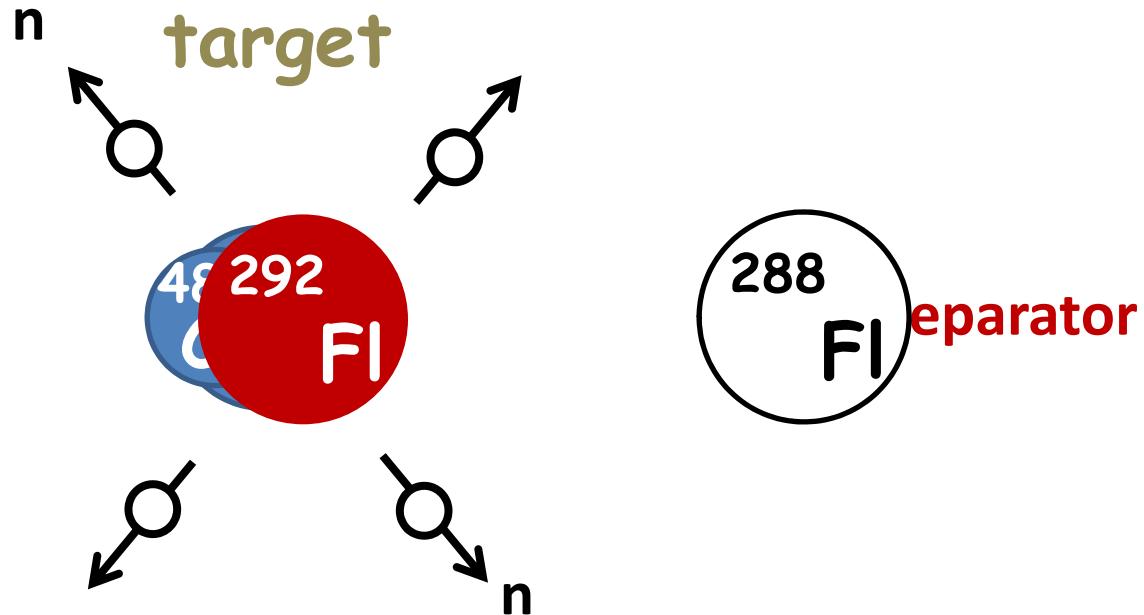


from
accelerator

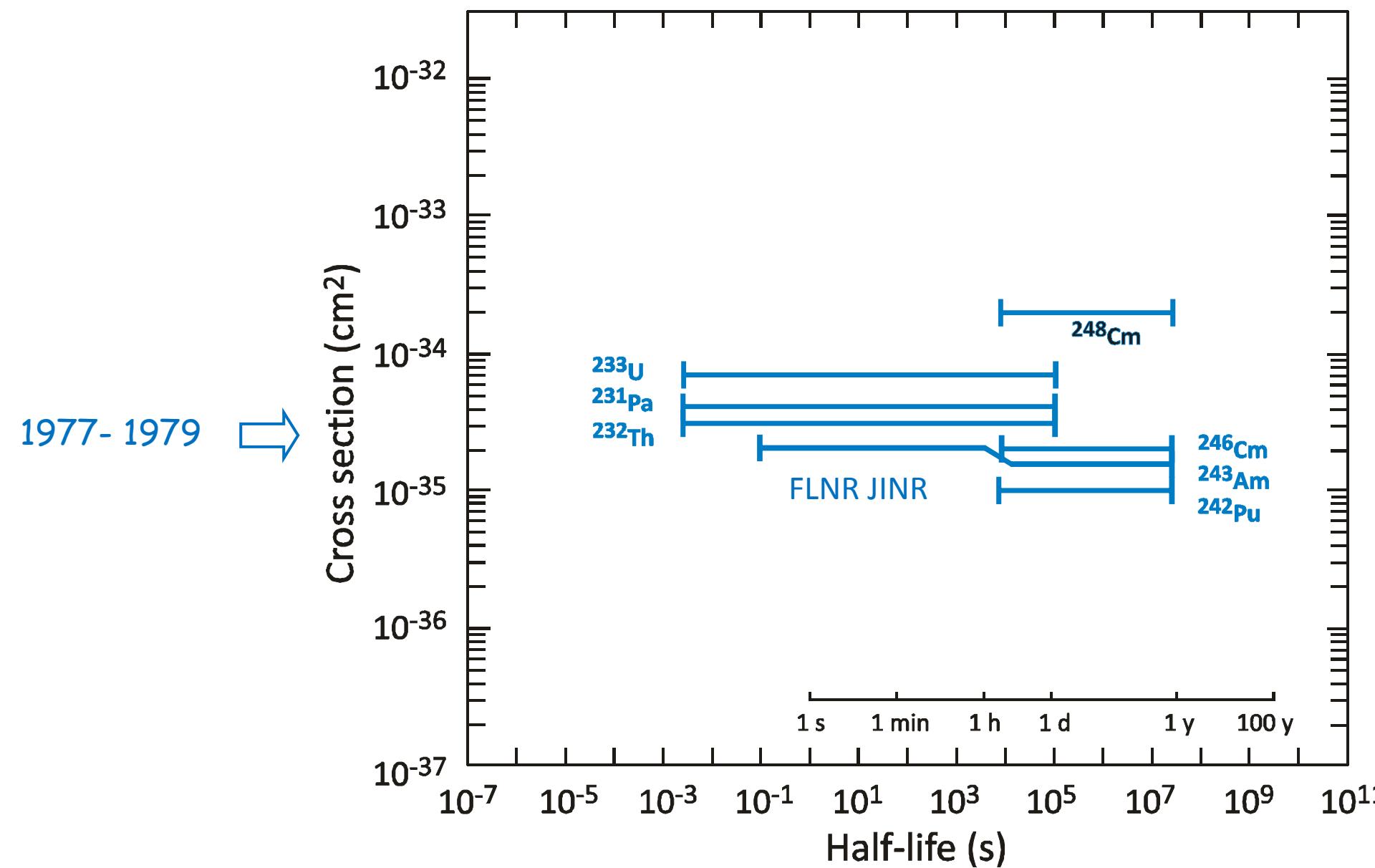
projectiles

rare and very
expensive isotope of Ca

artificial element from
flux nuclear reactor



Experiments on the synthesis of SHE with ^{48}Ca -beam at FLNR (JINR)

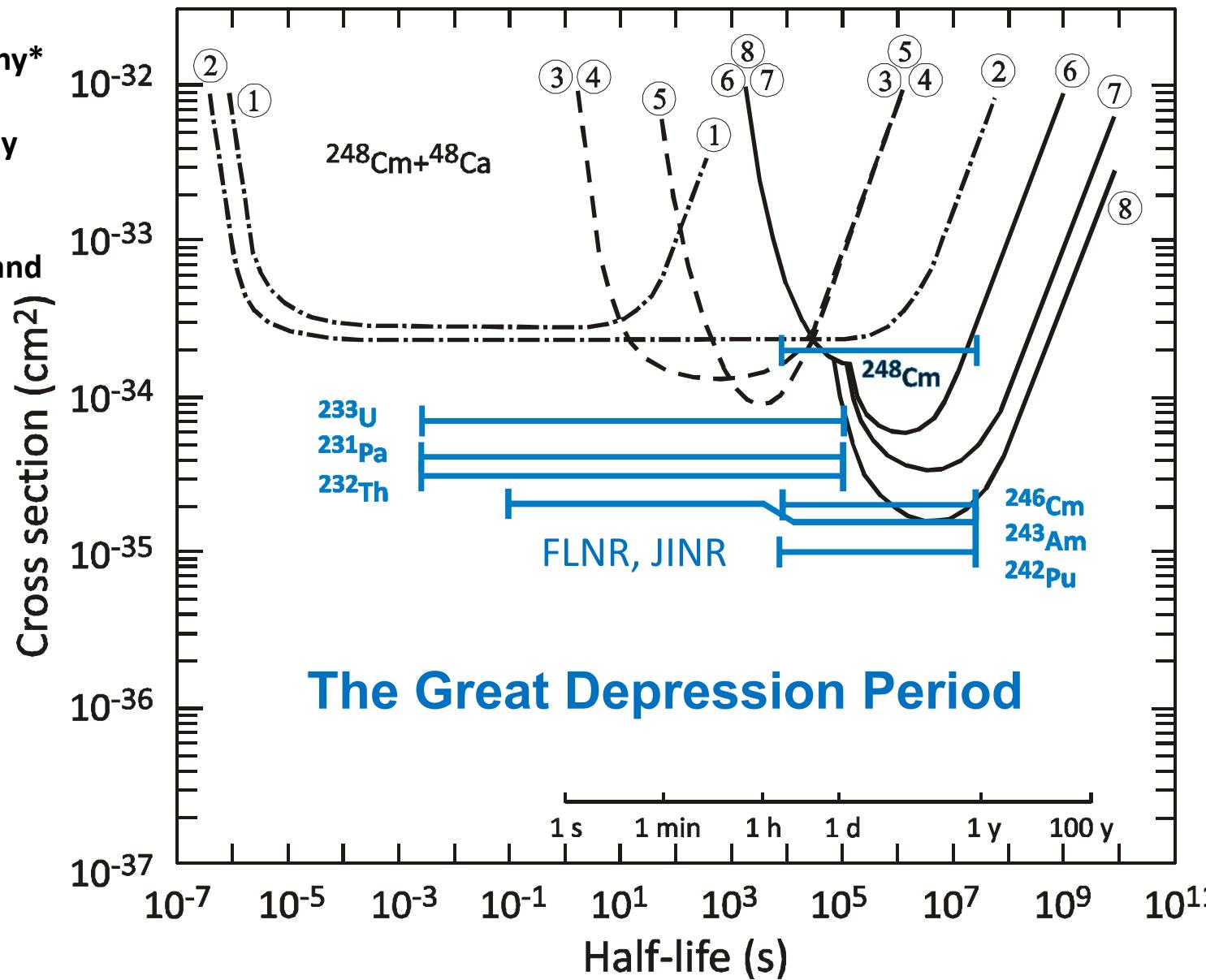


Experiments on the synthesis of Z=116 nuclei in the reaction $^{248}\text{Cm}+^{48}\text{Ca}$

GSI, Darmstadt, Germany*
 LBL, UC Berkeley, CA
 Univ. of Mainz, Germany
 LANL, Los Alamos, NM
 EIR, Würenlingen,
 Switzerland

1985 →

1977- 1979 →



Experiments on the synthesis of SHE in ^{48}Ca induced reaction

GSI, Darmstadt, Germany*

LBL, UC Berkeley, USA

Univ. of Mainz, Germany

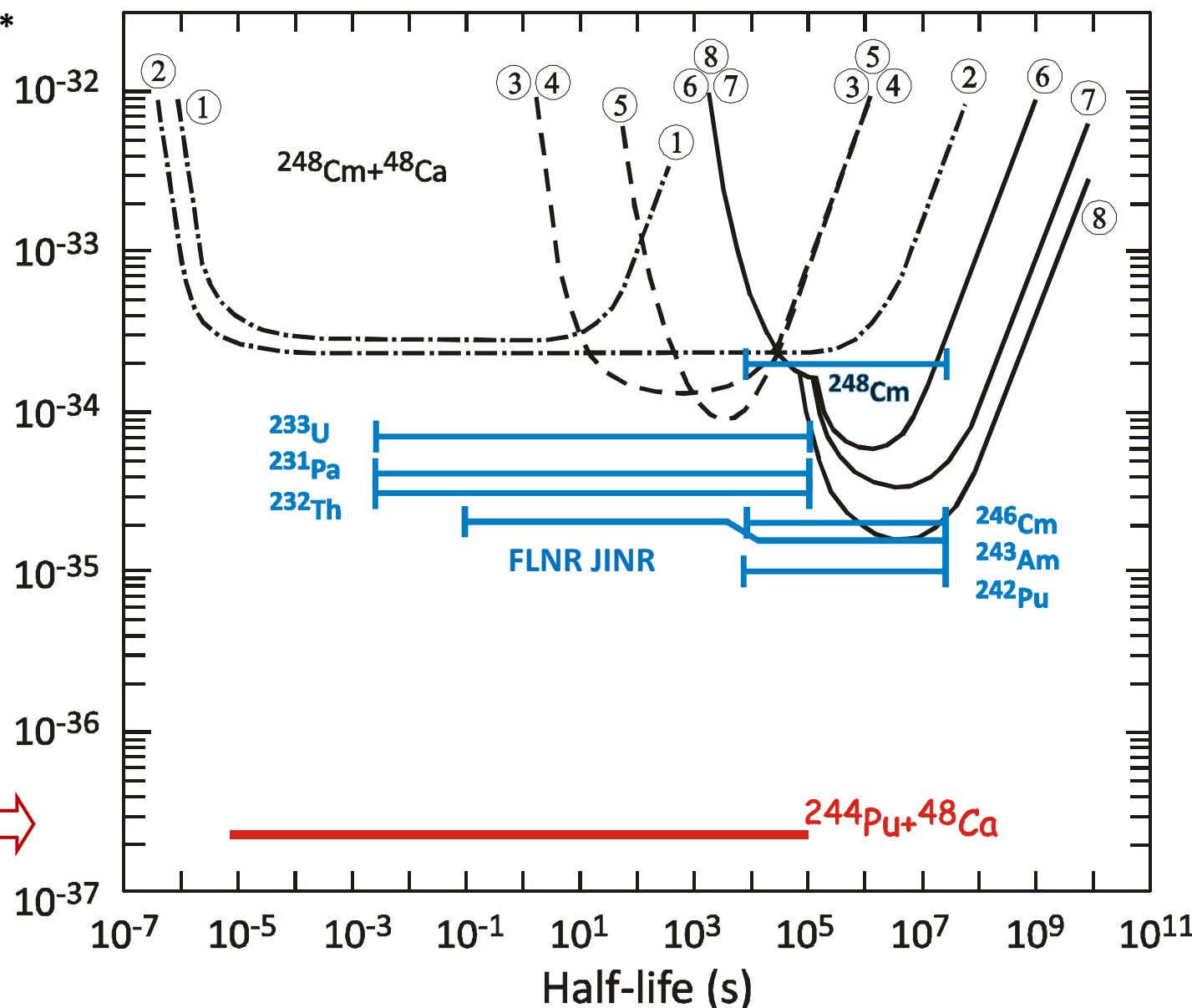
LANL, Los Alamos, USA

PSI, Villigen, Switzerland

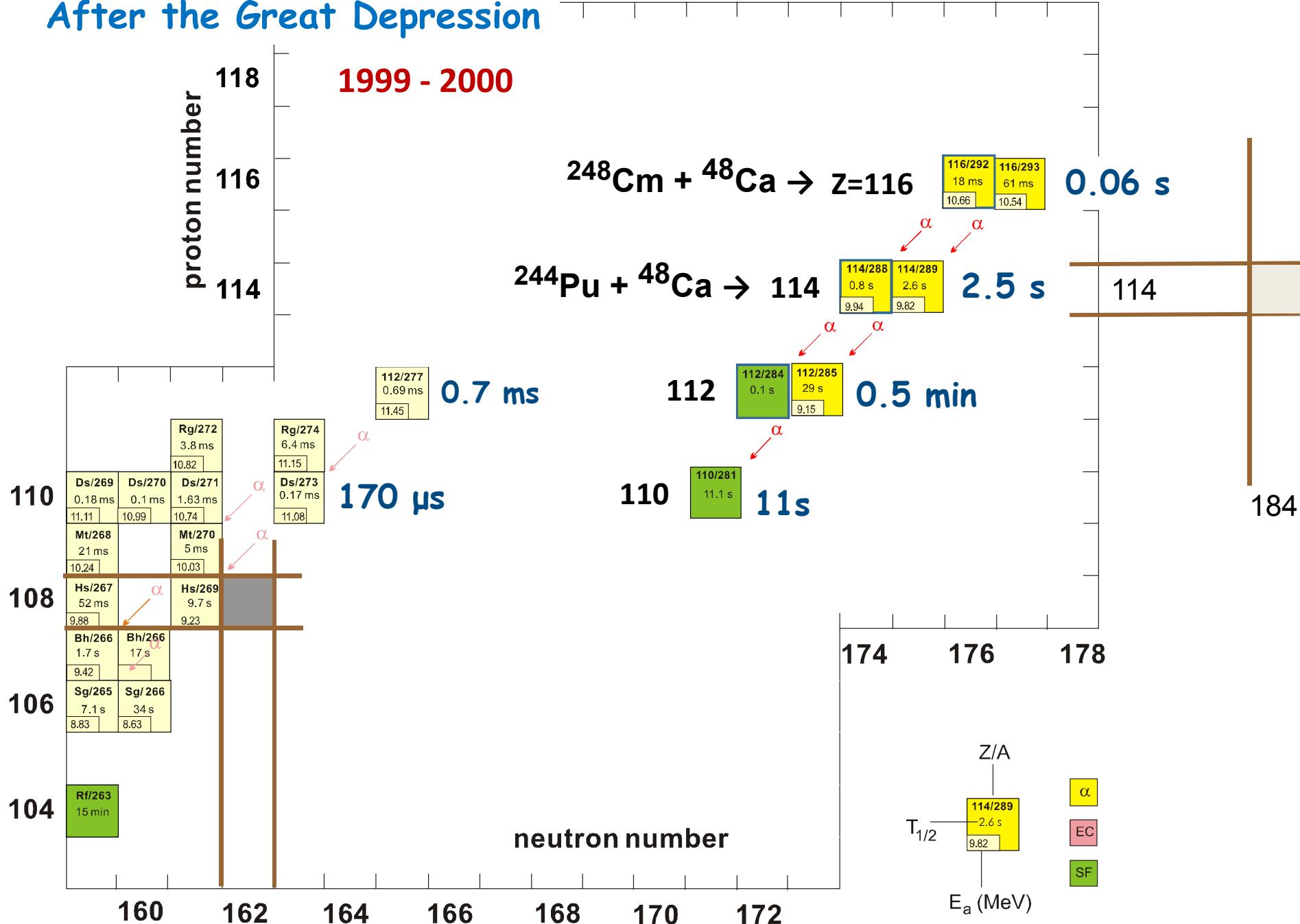
1985

1977- 1979

FLNR / LLNL 1999



After the Great Depression

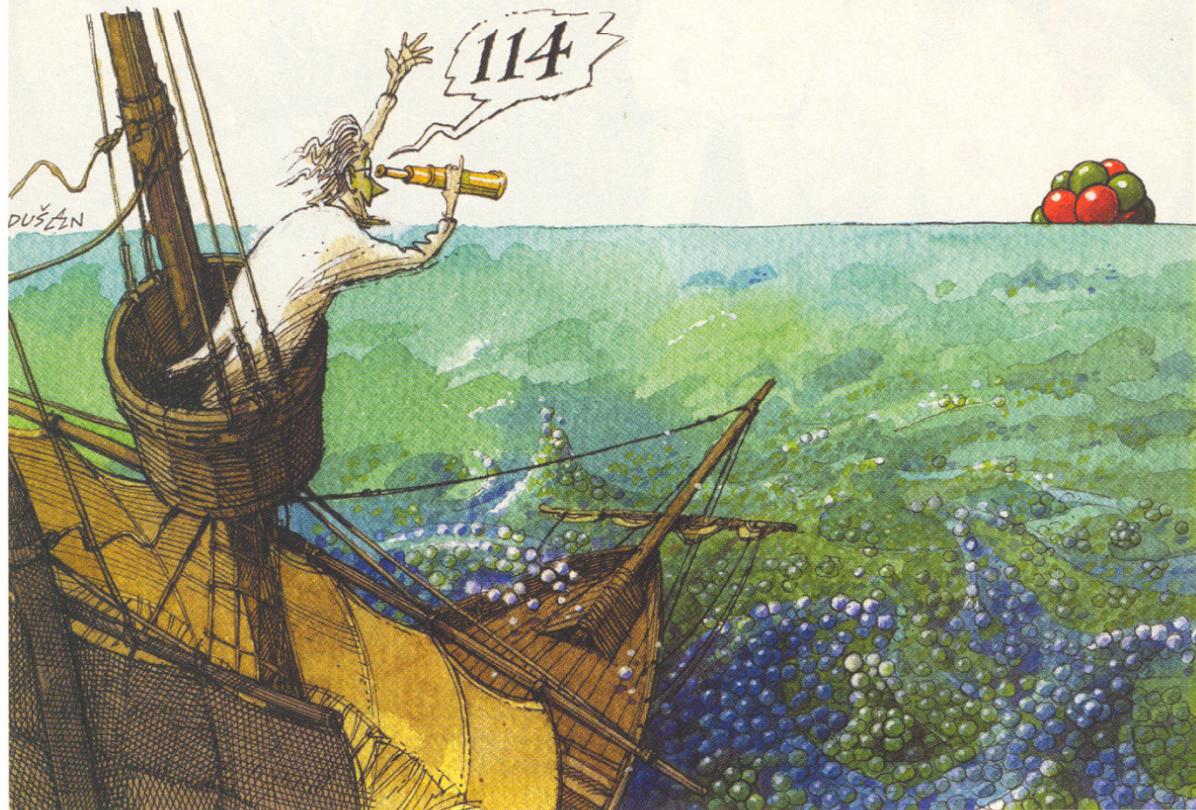


an Article from

SCIENTIFIC
AMERICAN

JANUARY 2000 VOL. 282 NO. 1

Voyage to SUPERHEAVY Island



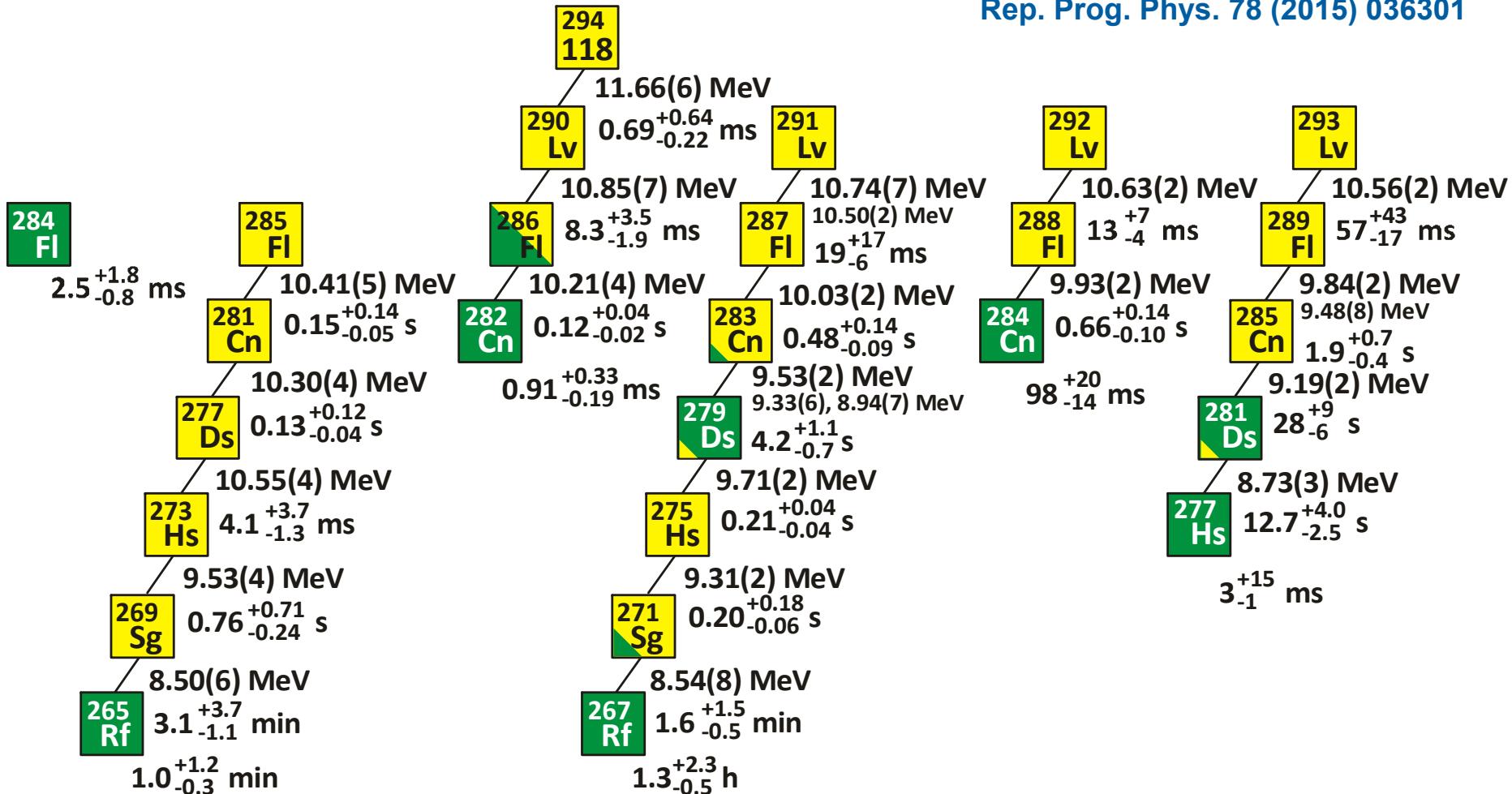
The synthesis of element 114 confirmed decades-old theoretical predictions of a little patch of nuclear stability in a sea of short-lived superheavy nuclei

Yuri Oganessian. Discovery of the Island of Stability for SHE. May 19, 2017, Copenhagen, Denmark

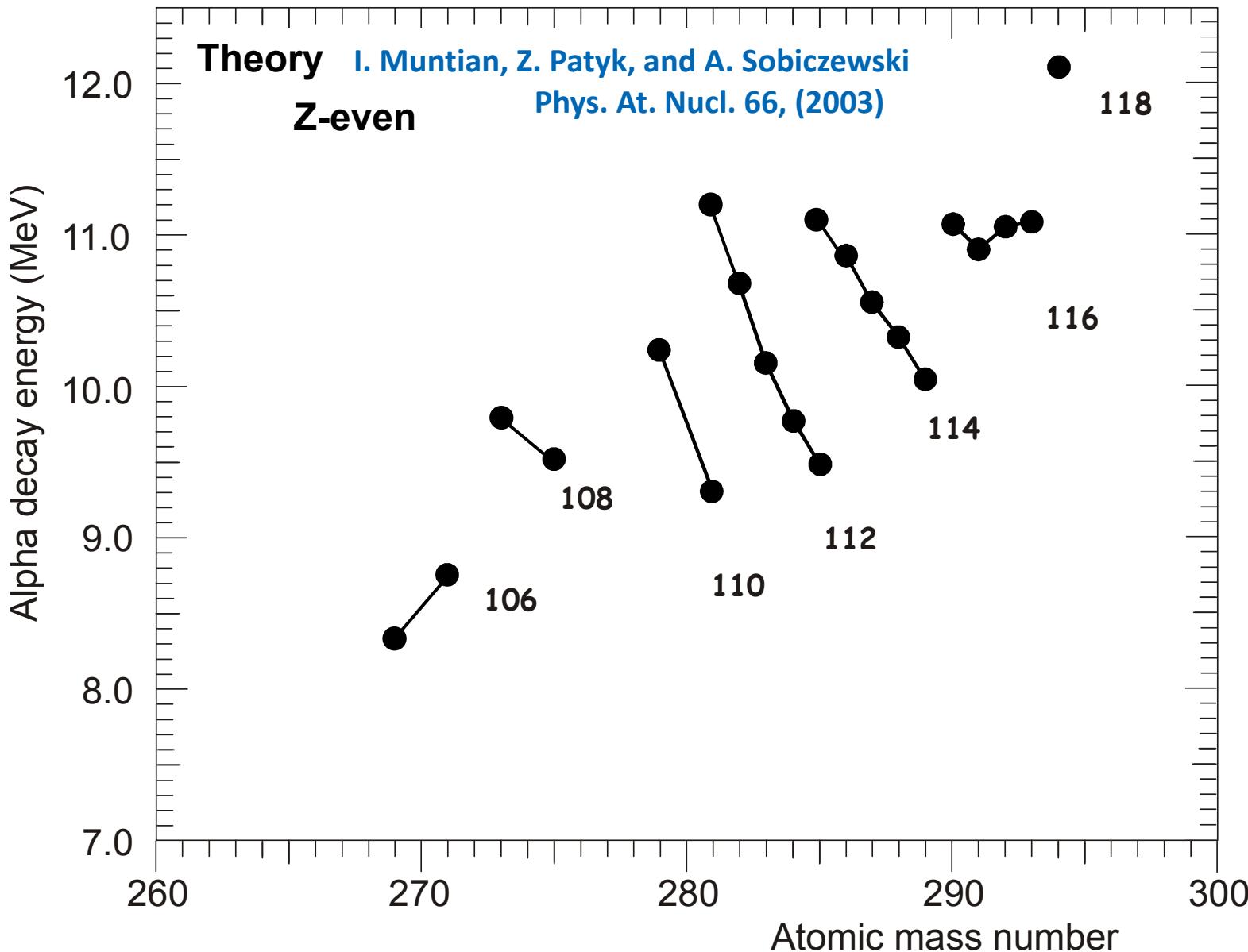
Summary decay properties of the isotopes of elements 112, 114, 116 and 118 observed in ^{238}U , $^{240,242,244}\text{Pu}$, $^{245,248}\text{Cm}$ and $^{249}\text{Cf} + ^{48}\text{Ca}$ reactions

2015

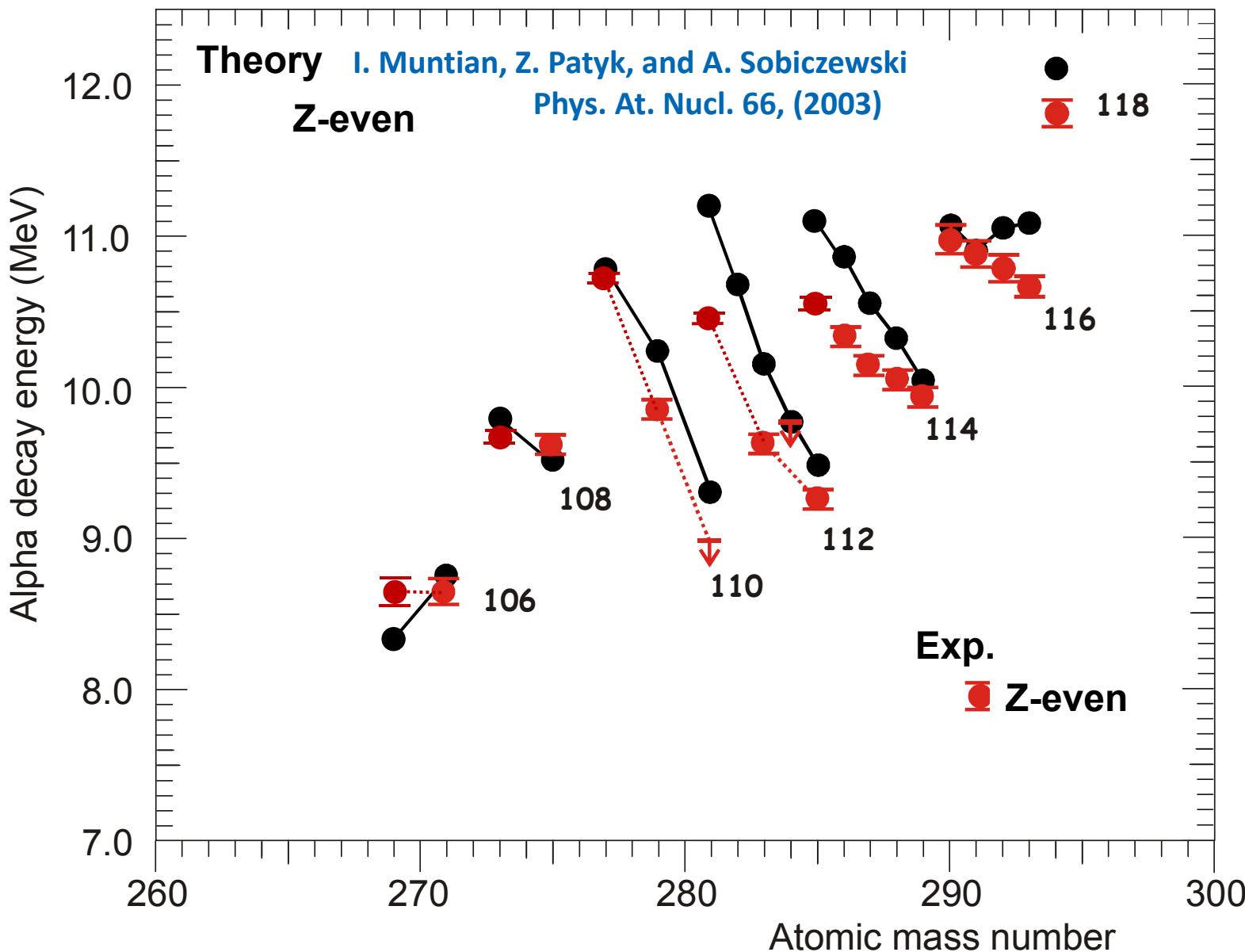
Yu Ts Oganessian and V K Utyonkov,
Rep. Prog. Phys. 78 (2015) 036301



Alpha - decay

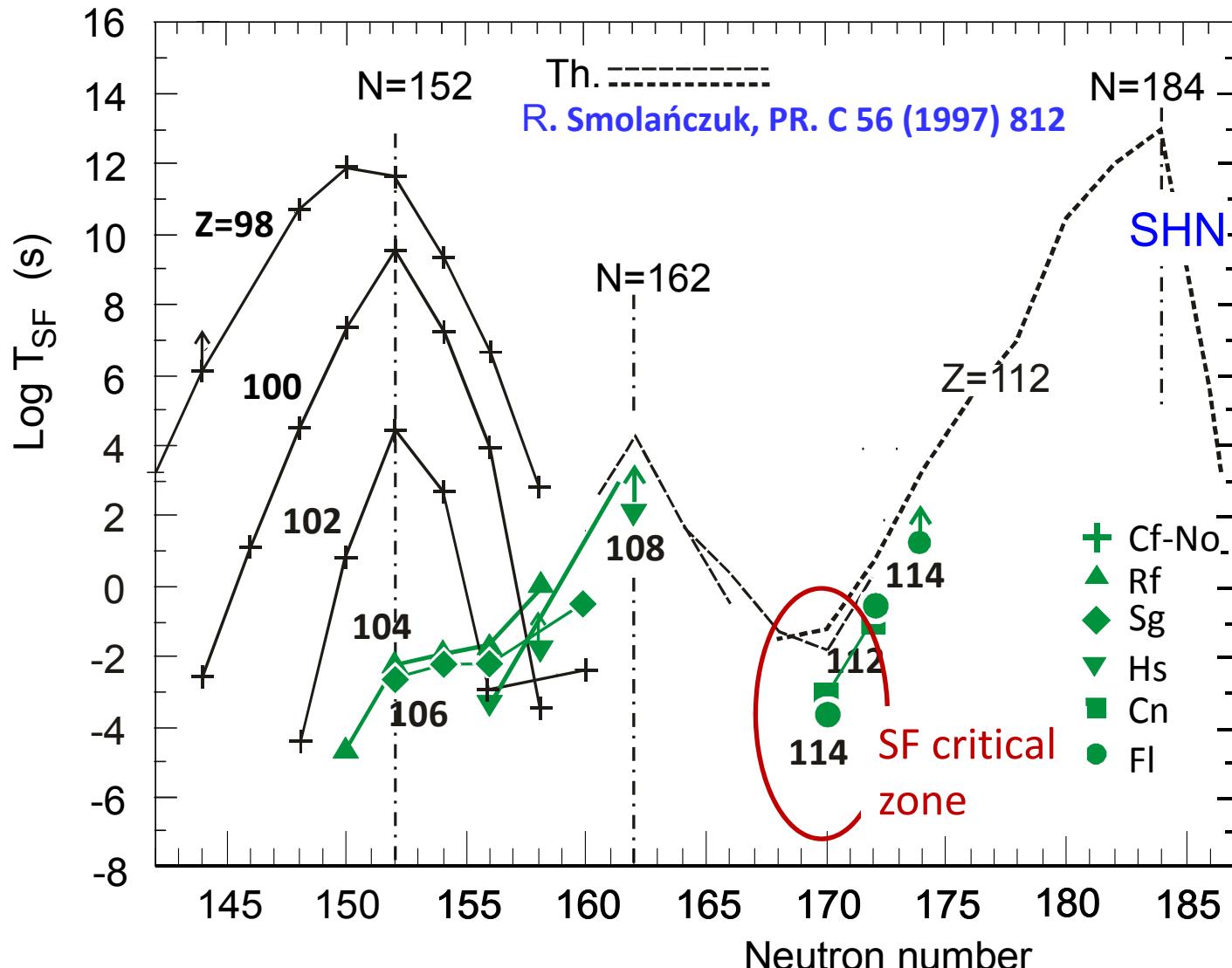


Alpha - decay



Spontaneous fission

even-even isotopes



Cross sections

hot fusion

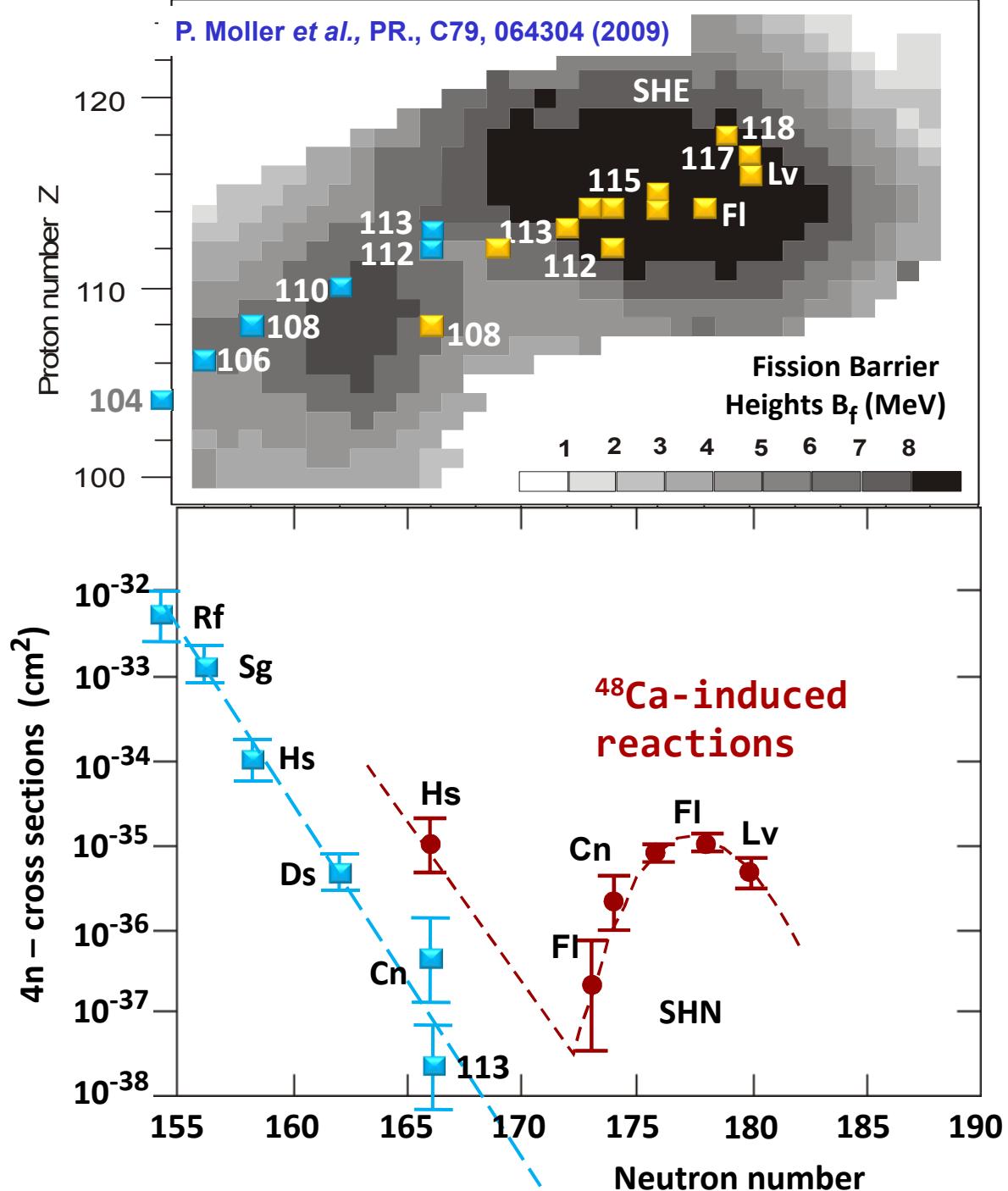
$^{48}\text{Ca} + ^{238}\text{U}, \dots, ^{249}\text{Cf}$

$E_x = 40 - 45 \text{ MeV}$
 $x = 4 - 5$

cold fusion

$^{208}\text{Pb}, ^{209}\text{Bi} + ^{50}\text{Ti}, \dots, ^{70}\text{Zn}$

$E_x = 12 - 15 \text{ MeV}$
 $x = 1$



Odd-Z Superheavy Nuclei

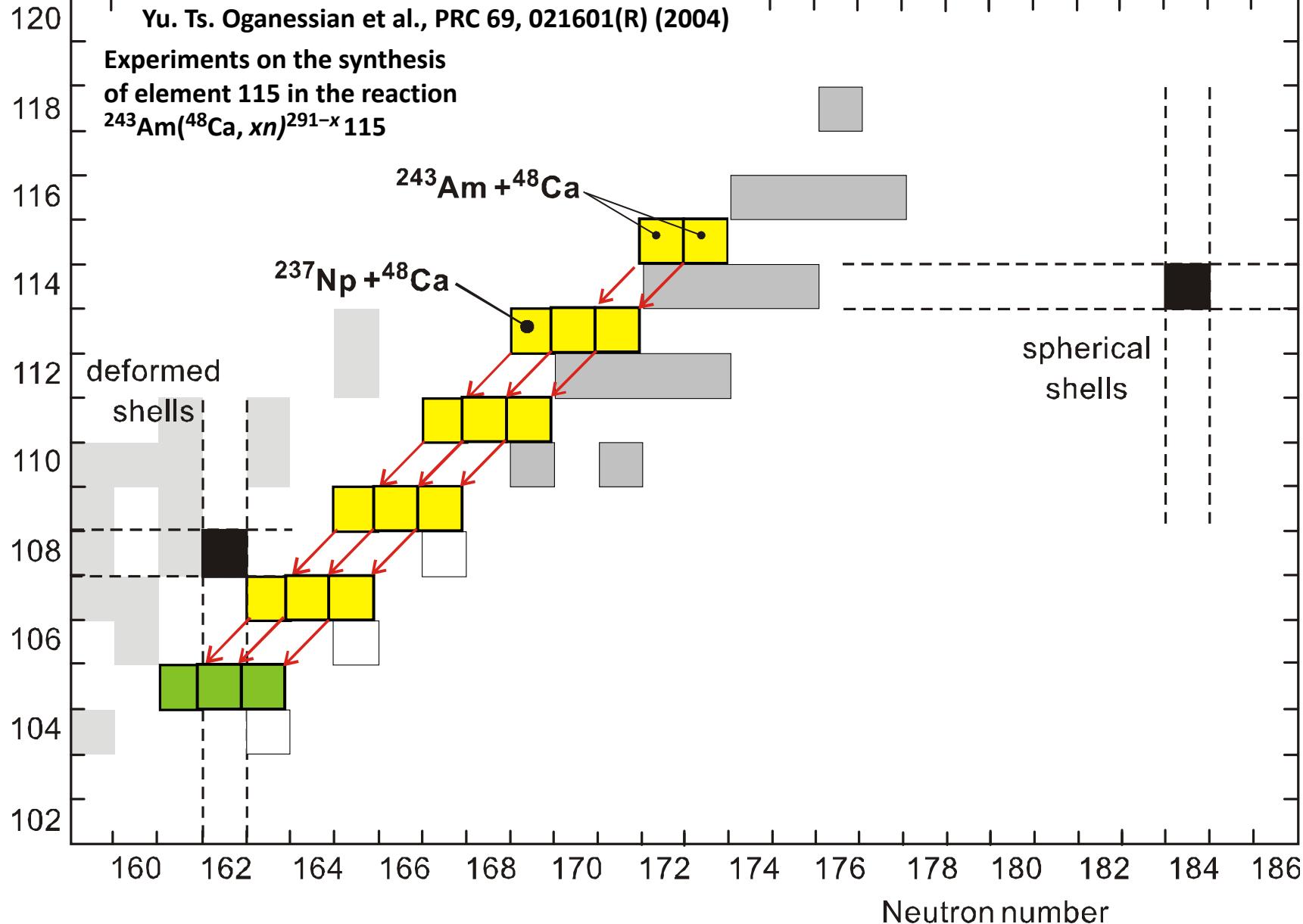
Synthesis of the Isotopes
with Z =113, 115 and 117

RAPID COMMUNICATIONS

Yu. Ts. Oganessian et al., PRC 69, 021601(R) (2004)

Experiments on the synthesis
of element 115 in the reaction
 $^{243}\text{Am}(^{48}\text{Ca}, xn)^{291-x} 115$

Proton number



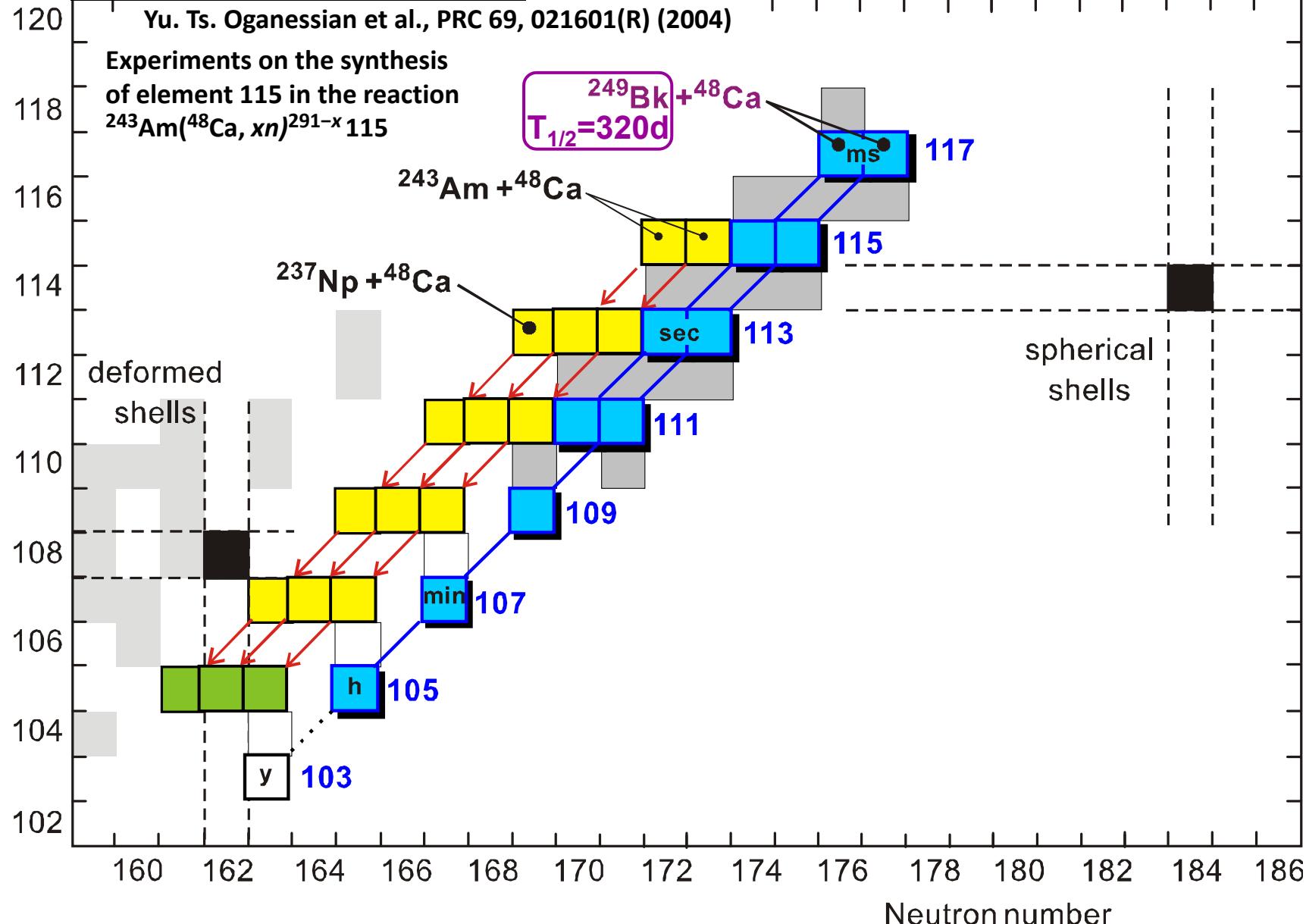
RAPID COMMUNICATIONS

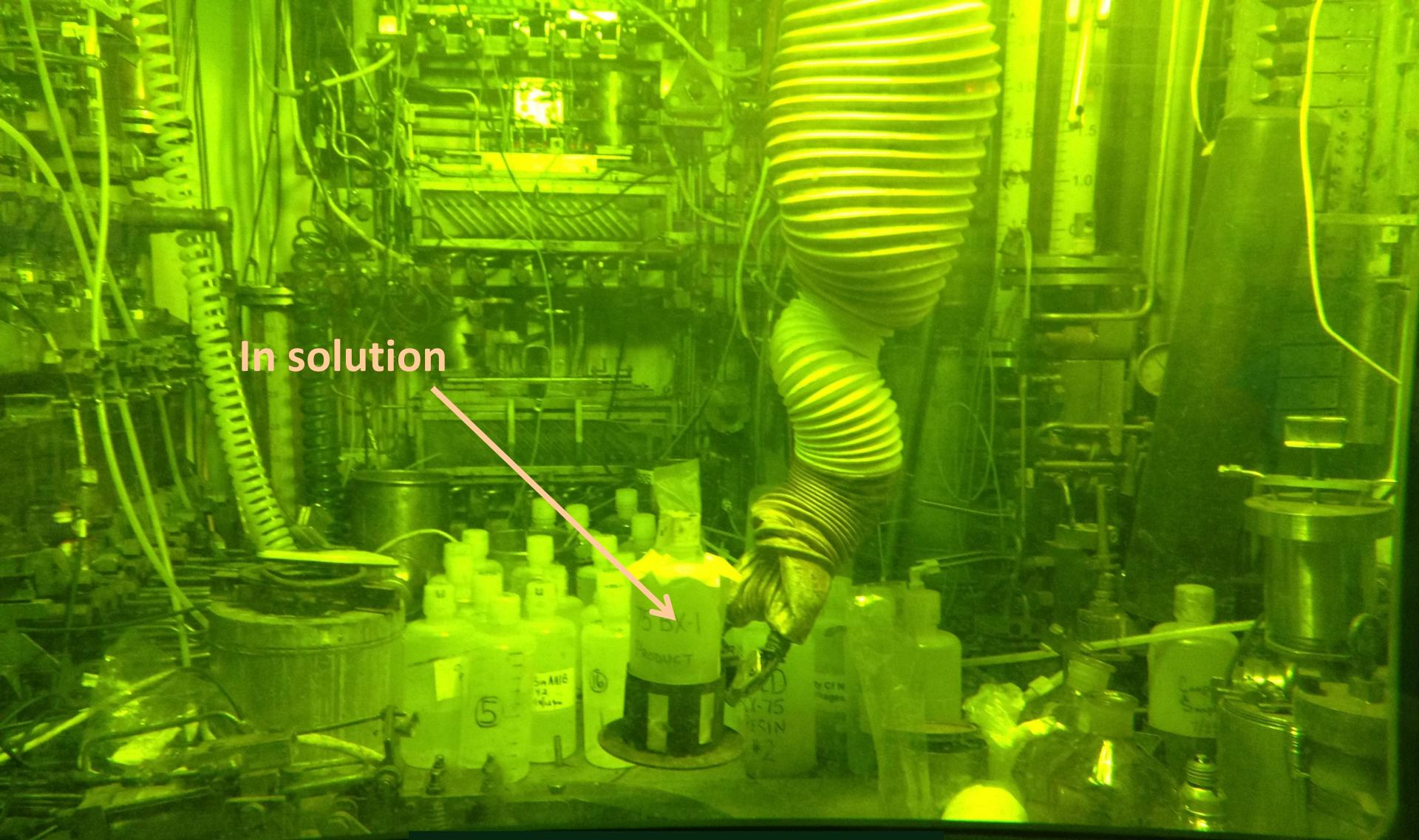
Yu. Ts. Oganessian et al., PRC 69, 021601(R) (2004)

Experiments on the synthesis
of element 115 in the reaction
 $^{243}\text{Am}(^{48}\text{Ca}, xn)^{291-x} 115$

$^{249}\text{Bk} + ^{48}\text{Ca}$
 $T_{1/2} = 320\text{d}$

Proton number





Berkelium -249 at hot cell

Yuri Oganessian. Discovery of the Island of Stability for SHE. May 19, 2017, Copenhagen, Denmark



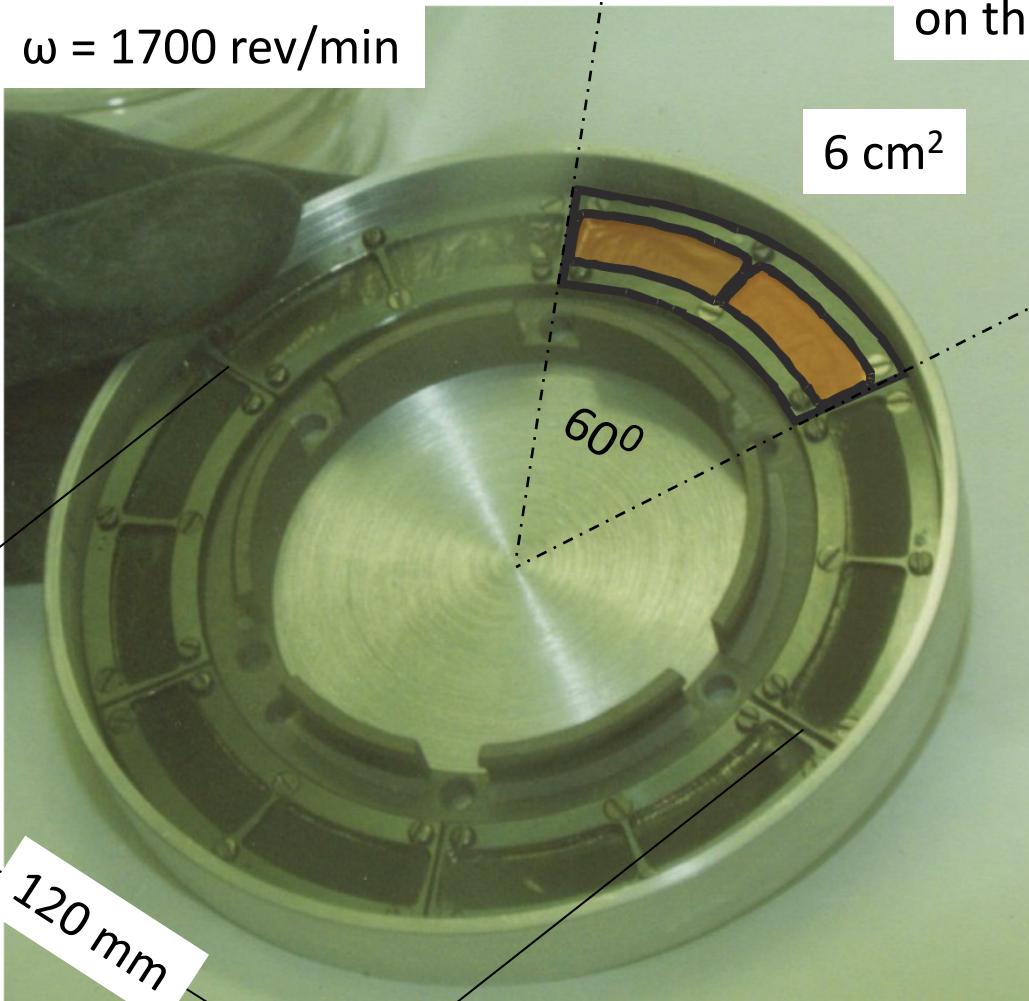
The purified product contained 22.2 μg Bk-249

Bk(NO₃)₃ Product

Target

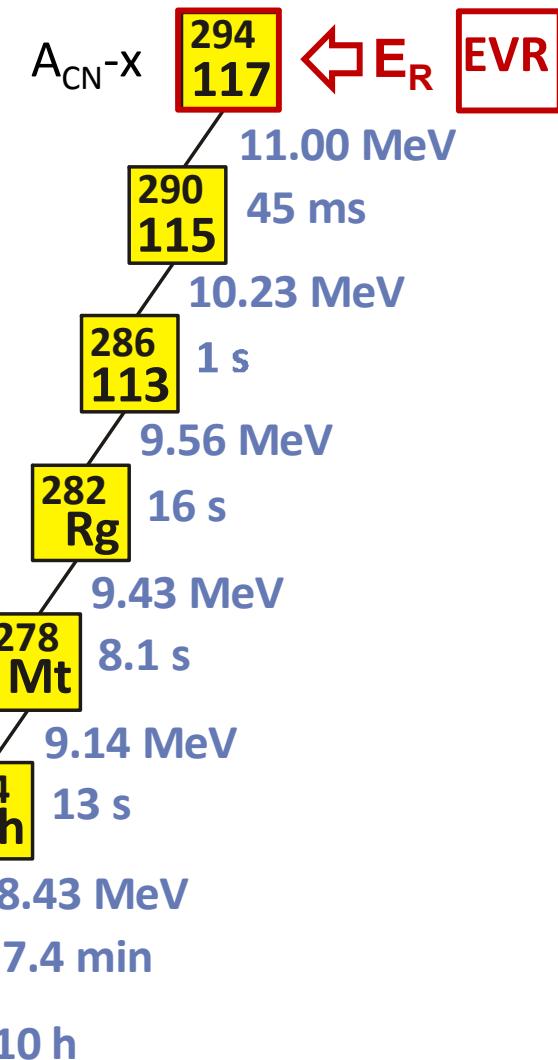
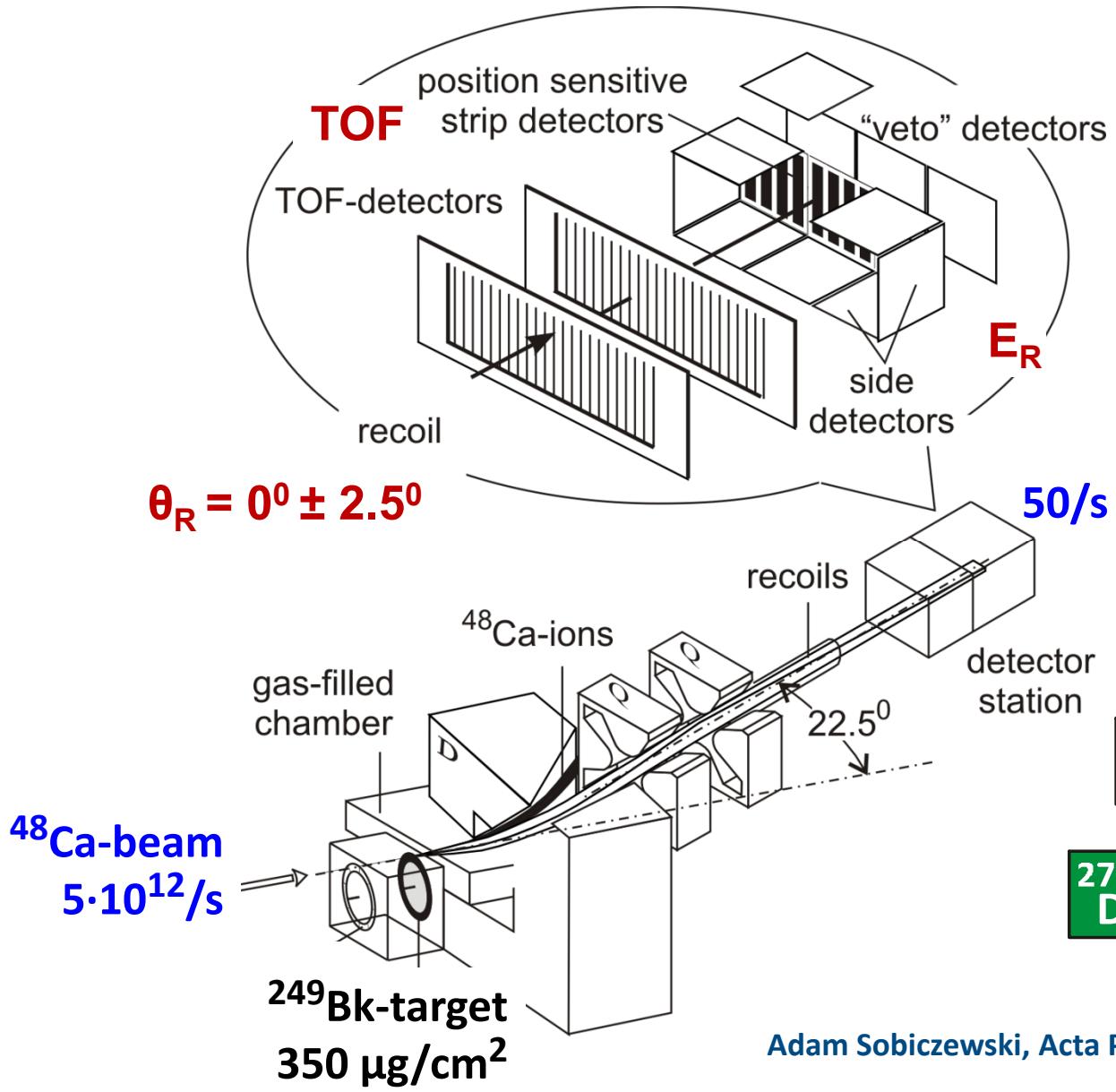
$\omega = 1700$ rev/min

350 mg/cm² deposited
on the 1.5 μm -Ti foil



Yu. Oganessian 2010

Dubna Gas-Filled Recoil Separator DGFRS



Adam Sobiczewski, Acta Phys. Pol. B 41, 157 (2010).



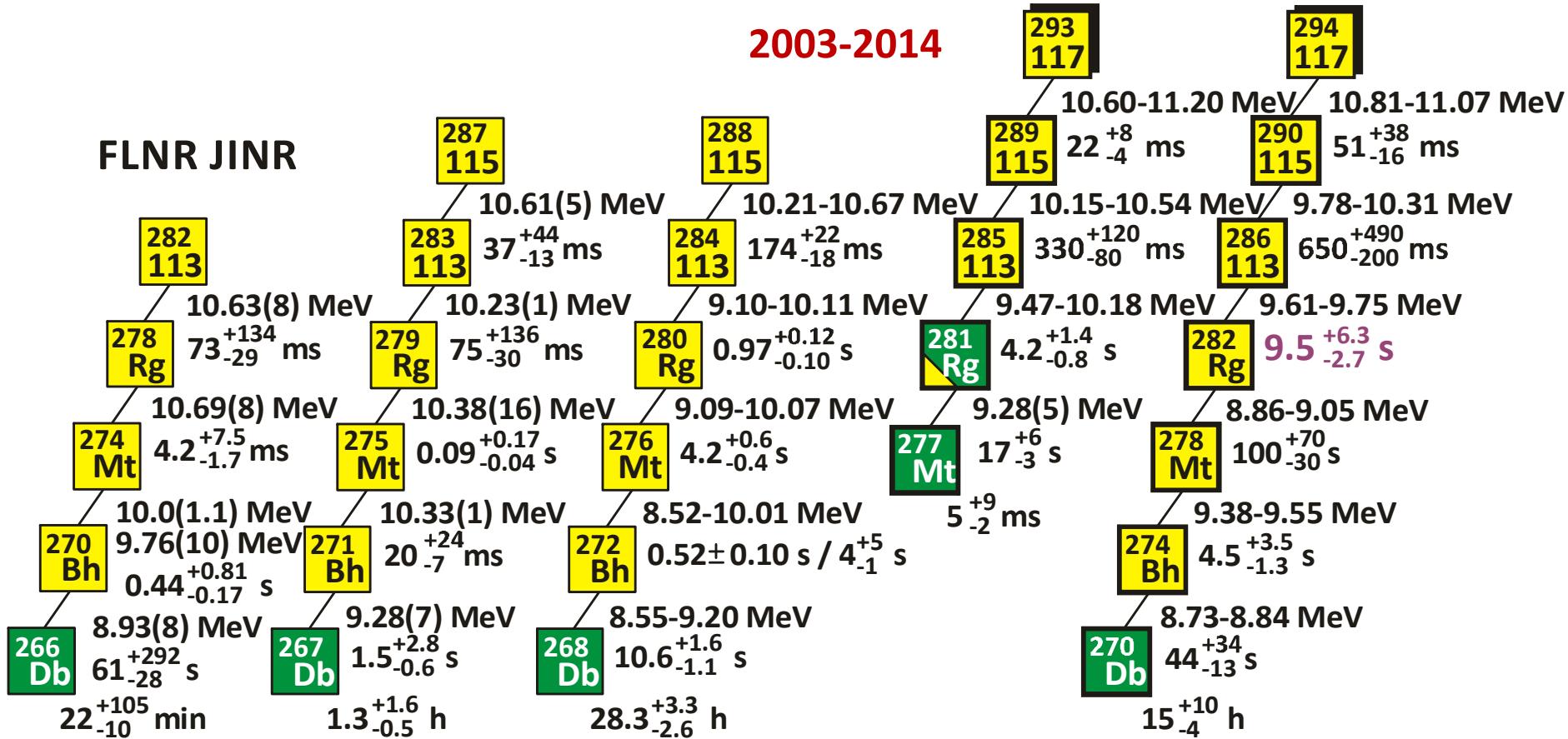
Heavy Ion Accelerator, FLNR

Yuri Oganessian. Discovery of the Island of Stability for SHE. May 19, 2017, Copenhagen, Denmark

Summary decay properties of the isotopes of elements 113, 115, and 117
 observed in ^{237}Np , ^{243}Am and $^{249}\text{Bk} + ^{48}\text{Ca}$ reactions

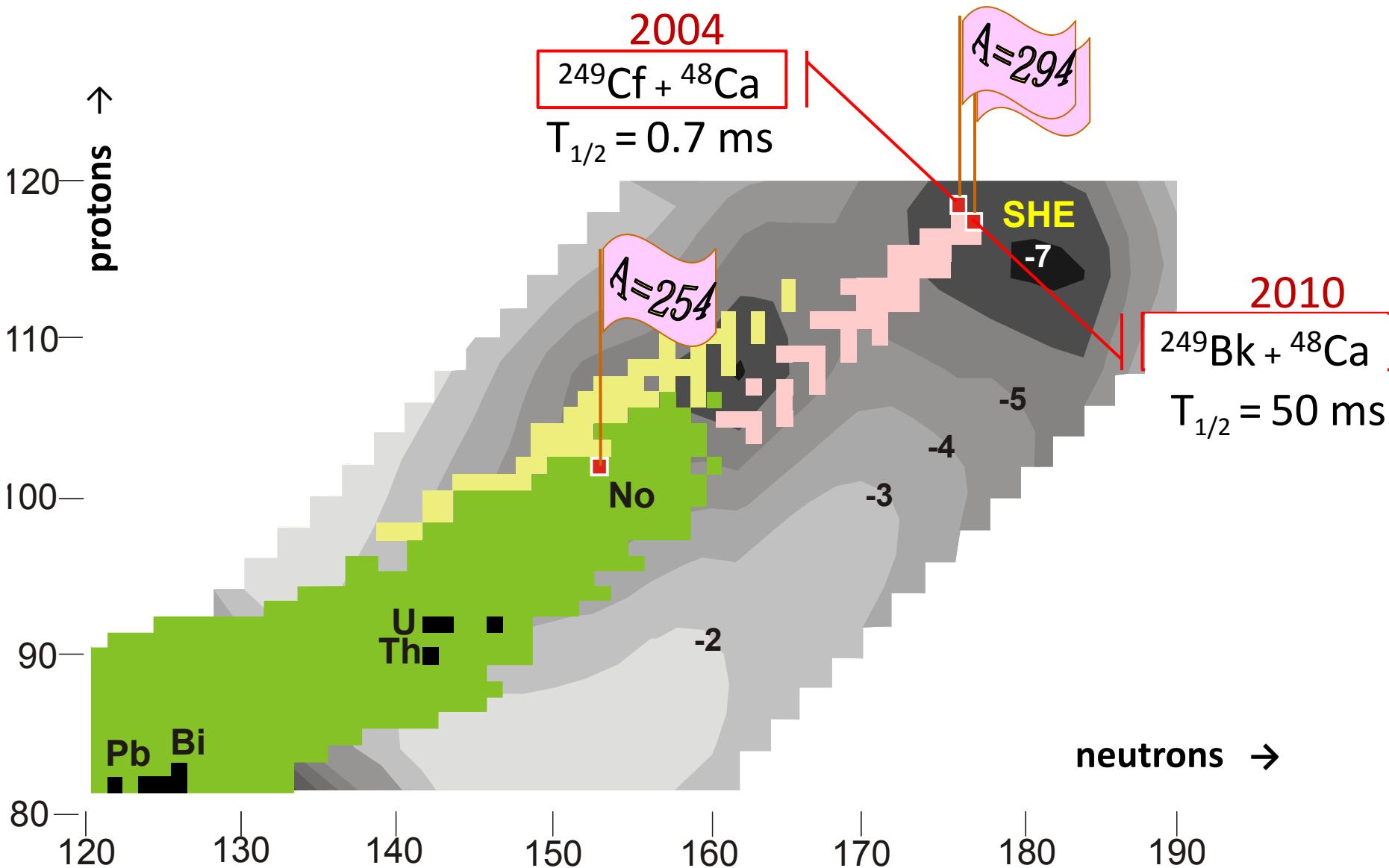
FLNR JINR

2003-2014



Discovery of the Heaviest Nuclei with Z=112-117 [Confirmations 2007-2014]

A/Z	Setup	Laboratory	Publications
$^{283}\text{112}$	SHIP	GSI Darmstadt	Eur. Phys. J. A32, 251 (2007)
$^{283}\text{112}$	COLD	PSI-FLNR (JINR)	NATURE 447, 72 (2007)
$^{286, 287}\text{114}$	BGS	BNL (Berkeley)	P.R. Lett. 103, 132502 (2009)
$^{288, 289}\text{114}$	TASCA	GSI – Mainz	P.R. Lett. 104, 252701 (2010)
$^{292, 293}\text{116}$	SHIP	GSI Darmstadt	Eur. Phys. J. A48, 62 (2012)
$^{287, 288}\text{115}$	TASCA	GSI – Mainz	P.R. Lett. 111, 112502 (2013)
$^{293, 294}\text{117}$	TASCA	GSI – Mainz	P.R. Lett. 112, 172501 (2014)
$^{292, 293}\text{116}$	GARIS	RIKEN Tokyo	Accelerator Progress Rep. (2013)



With $Z > 40\%$ larger than that of Bi, we see an impressive extension in nuclear survivability.

Although SHN are at the limits of Coulomb stability,

- shell stabilization lowers ground-state energy,
- creates a fission barrier,
- and thereby enables SHN to exist.

The fundamentals of the modern theory concerning the mass limits of nuclear matter have obtained experimental verification

Super Heavy Atoms

Chemistry of the SHE

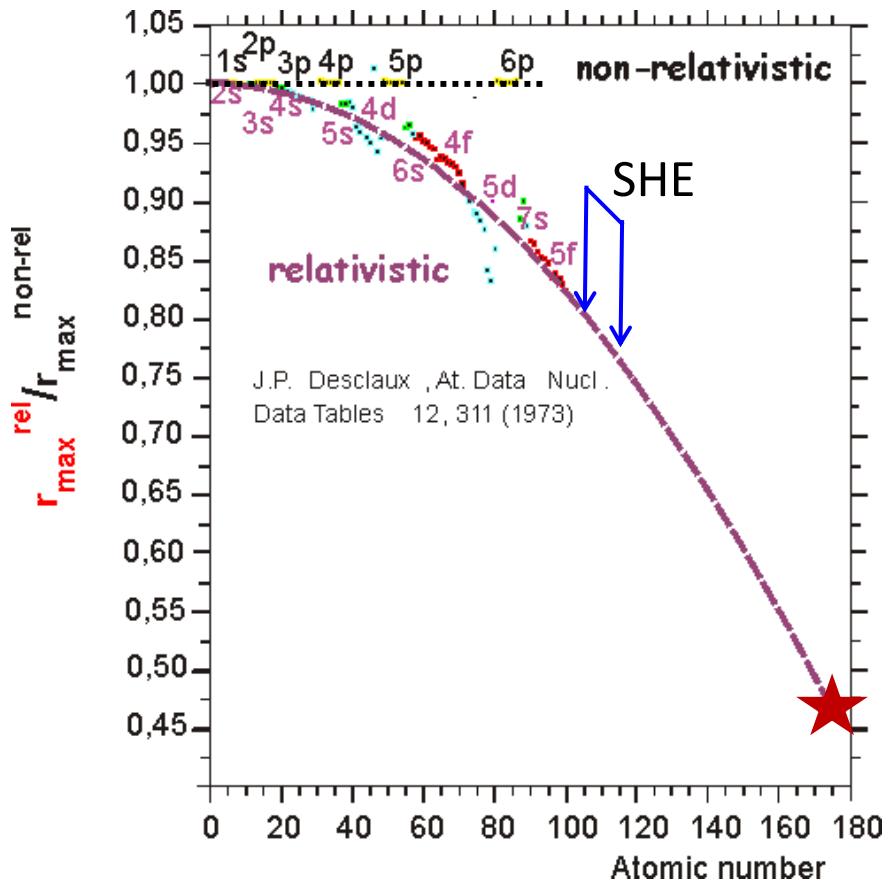


Table of Elements
от Z=1 до Z=172

Period 1																		18 Orbitals																	
1	H	2	13	14	15	16	17	2	He	1s																									
2	Li	Be	5	6	7	8	9	3	2s2p																										
3	Na	Mg	13	14	15	16	17	10	3s3p																										
4	K	Ca	13	14	15	16	17	11	4s3d4p																										
5	Rb	Sr	13	14	15	16	17	12	5s4d5p																										
6	Cs	Ba	13	14	15	16	17	13	6s5d6p																										
7	Fr	Ra	104	105	106	107	108	14	7s6d7p																										
8	119	120	109	110	111	112	113	15	8s7d8p																										
9	165	166	114	115	116	117	118	16	9s9p																										
			104	105	106	107	108	17																											
			109	110	111	112	113	18																											
			114	115	116	117	118	19																											
			121	122	123	124	125	126	20																										
			127	128	129	130	131	132	21																										
			133	134	135	136	137	138	22																										
			139	140	141	142	143	144	23																										
			145	146	147	148	149	150	24																										
			151	152	153	154	155	156	25																										
			157	158	159	160	161	162	163	26																									
			164	165	166	167	168	169	170	27																									
			171	172						28																									

LANTANIDES

ACTINIDES

4f

5f

6f

5g

Relativistic Contraction

Calculated by P. Dirac and V. Fock
in non-relativistic approximation

Period 1

1	H	2
2	3 Li	4 Be
3	11 Na	12 Mg
4	19 K	20 Ca
5	37 Rb	38 Sr
6	55 Cs	56 Ba
7	87 Fr	88 Ra
8	119	120

Periodic Table Z=1-138

Calculated in non-relativistic
approximation of Dirac - Fok

13	14	15	16	17	18	Orbitals									
5 B	6 C	7 N	8 O	9 F	10 Ne	1s									
13 Al	14 Si	15 P	16 S	17 Cl	Ar	2s2p									
31 Ga	32 Ge	33 As	34 Se	35 Br	Kr	3s3p									
49 In	50 Sn	51 Sb	52 Te	53 I	Xe	4s3d4p									
81 Tl	82 Pb	83 Bi	84 Po	85 At	Rn	5s4d5p									
104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Ms	116 Lv	117 Ts	118 Og	7s6d7p

Completion of the 7-th row

121-

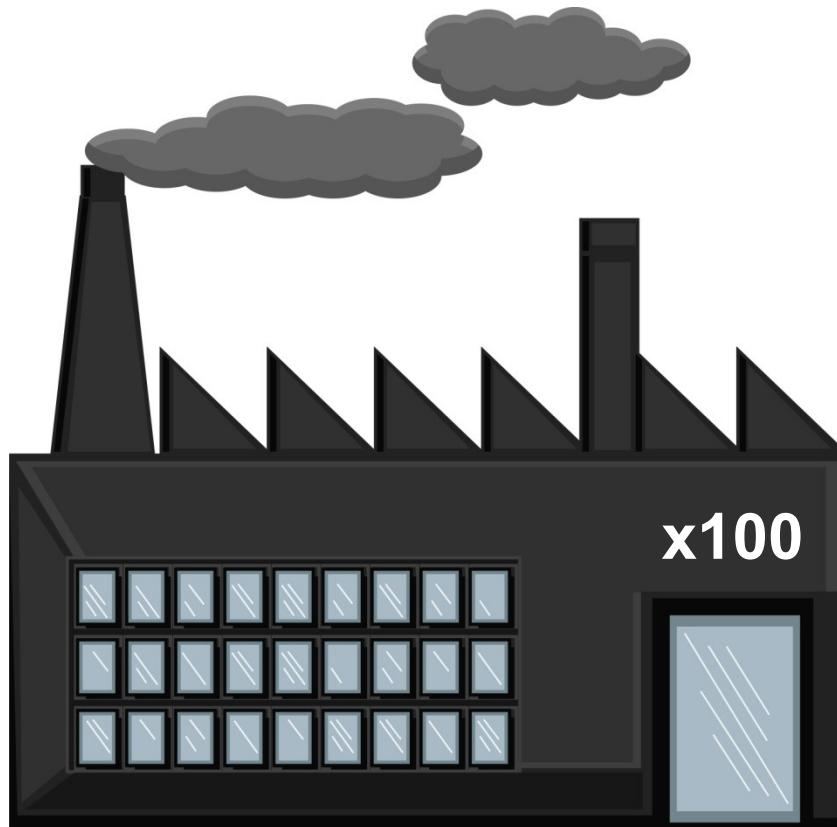


Is Element 118 – noble gas?

8	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	5g
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What's next?

In order to expand the field of research it is necessary first of all to increase the sensitivity of experiments



SHE-Factory

SHE-Factory joining of efforts

Isotope production:
Cm-248
Bk-249
Cf-251

To be increased
10 times

New accelerator
High beam
dose of : Ca-48
Factor 10-20 Ti-50
Ni-64

Depend of
target durability

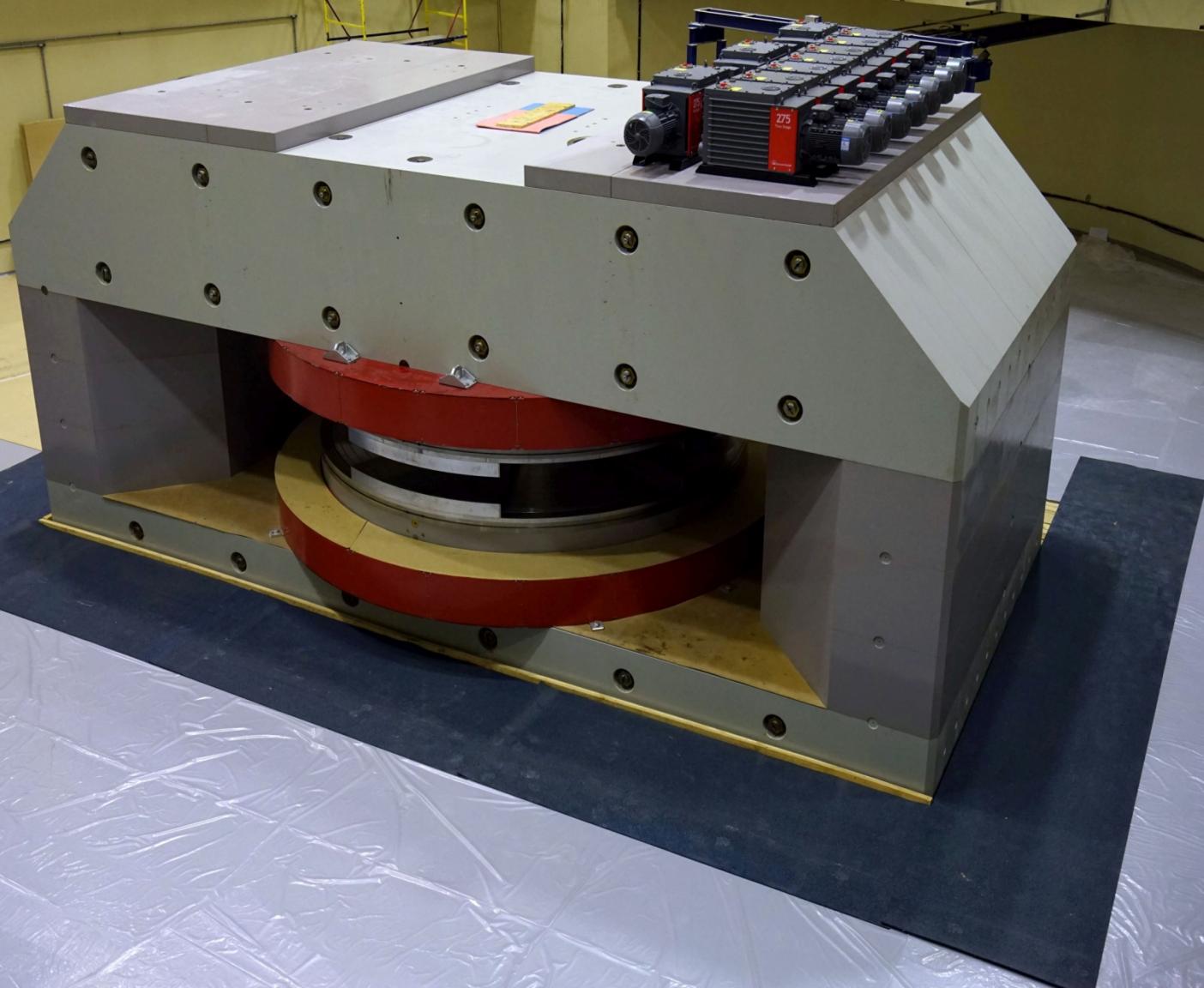
SC- recoil separator
equipped with
Gas Catcher
On-line separator
& sophisticated
Detectors

Factor 3-5

is closely linked
to the intellect

August 2014, Dubna





Yuri Oganessian. Discovery of the Island of Stability for SHE. May 19, 2017, Copenhagen, Denmark

Collaboration

Thank you, very much!

