

Plasma Modified Production of High-current, High-purity cw H^+ , D^+ , and H^- Beams from Microwave-driven Sources*

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Abstract

We have recently reported [1] the production of cw proton beams from magnetically-confined microwave-driven sources, operating under non-resonant (non-ECR) conditions, with proton fractions greater than 0.95, the remaining fraction consisting of H_2^+ (0.05) with no H_3^+ . We achieve this by the addition of H_2O to the plasma at molecular concentrations of $\sim 1.0\%$ and about 700 W 2.45 GHz RF power to the source. High-current (45 mA) high-power (45 kV) beams of $>92\%$ proton purity have been produced using this technique [2]. Additional impurity ions, O^+ at 4 parts per thousand (ppt) and OH^+ and H_2O^+ at $\ll 1$ ppt are produced. We report further progress using this technique and similar results we have achieved for cw D^+ beams with D_2O and H_2O additives. Finally, we report progress we have made in the direct extraction of cw H^- beams from microwave-driven sources in terms of ion source surface material and confining magnetic field configurations. Mechanisms are discussed.

Introduction

High-current cw proton sources of high reliability are a current requirement for several proposed accelerator applications, including spallation neutron sources and accelerator production of tritium. A desirable property of such sources is that the proton fraction of the extracted beam be as high as possible so as to avoid the need for selection of the desired ion, i.e., to enable direct injection into an accelerating structure. A number of sources have been described in the literature that yield proton fractions on the order of 80% of the extracted beam, the other unwanted components being H_2^+ and H_3^+ . [3,4,5] Techniques to enhance the proton fraction above 80% are highly desirable.

Similarly, high current H^- sources are a requirement for advanced spallation sources based on circular accelerators and other circular proton accelerators that use stripping injection, and for tandem accelerators.

Background

It has long been known that addition of minor constituents to microwave-generated plasmas can modify the species composition of the plasma and can increase atomic neutral fractions relative to molecular species. Systematic studies of atomic hydrogen fractions from microwave-driven plasmas

have produced atomic hydrogen beams with close to 100% purity, [6] leading us to recognize that this may be a viable technique for the production of high purity ion beams.

In the case of H^- production, pioneering studies by Hall et al. [7] demonstrated the effectiveness of freshly evaporated tantalum surfaces in producing copious quantities of vibrationally excited molecular hydrogen (a required precursor to H^- formation). The observations of Hall et al. appear to never have been deliberately, or successfully, applied to the production of high-current cw H^- beams.

Apparatus, Diagnostics, and Experimental Techniques

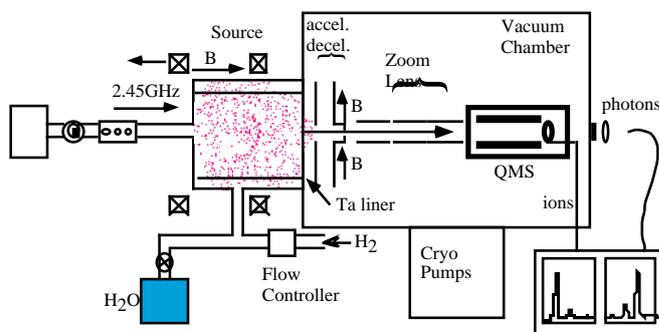


Figure 1: Schematic of apparatus including modifications necessary for production of H^- .

The major components of our apparatus shown in Fig. 1 include a magnetically-confined microwave-driven (ECR) source purchased from AECL powered by a 2.45 GHz microwave generator (2 kW). The microwave generator is coupled to the source via a circulator and a four-stub autotuner. The ion source is attached to a large high-vacuum oil-free diagnostic chamber with a base pressure of 1×10^{-8} Torr.

Ion beams extracted at a few hundred volts from the 5 mm source aperture by an accel-decel arrangement are primarily collected on the decel electrode that is in the configuration of a faraday cup. A 0.5 mm aperture in the decel electrode allows a small portion of the beam to be transported to a quadrupole mass spectrometer (QMS) via an electrostatic zoom lens for quantitative beam composition measurement. Light from the ion source is monitored by an optical monochromator by a clear line-of-sight through the QMS (sapphire window). The monochromator continuously monitors the atomic hydrogen

Balmer radiation (656 nm) to give a measure of the atom concentration in the source under varying conditions.

Additives, in this case H₂O and D₂O (gas), are introduced into the source along with the H₂ or D₂ via a micrometer leak valve. The composition of gas in the source (H₂, D₂, H₂O, D₂O, etc.) is monitored by a residual gas analyzer (RGA) mounted in the diagnostic chamber.

Results

Most of the measurements reported here are conducted with the ECR source operating slightly off-resonance, as chosen by the magnitude of the magnetic confining field. Although operation off-resonance usually produces a smaller fraction of protons in the extracted beam, such is not the case in the present experiments with H₂O as an additive.

A) Proton production

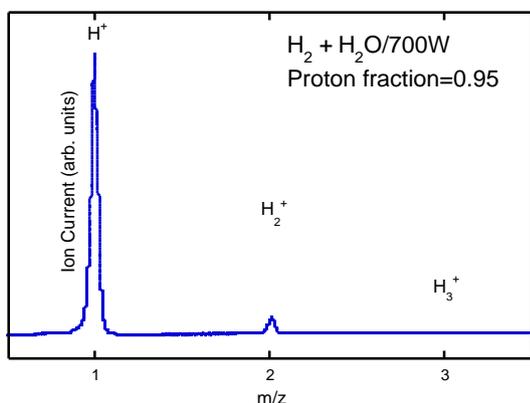


Figure 2: Mass spectrum of positive ion beam with H₂O additive.

Figure 2 shows a beam composition obtained with ~1% H₂O added to the source for 1 sccm H₂ flow and 700 W microwave power. In this case, the proton fraction of the total beam (consisting of H⁺ + H₂⁺) is slightly higher than 0.95. Under similar conditions with no H₂O, the proton fraction of the total beam (consisting of H⁺ + H₂⁺ + H₃⁺) is about 0.75, consistent with the measurements of Taylor and Wills [3] obtained under similar non-resonant conditions.

Figure 3 shows the proton fraction (H⁺/(H⁺+H₂⁺+H₂⁺)) and H atom emission from the source versus percent H₂O added to the plasma. As expected, the proton fraction extracted tracks the concentration of H atoms in the source. The hysteresis in the two sets of curves occurs because the measurements were taken under increasing and decreasing amounts of H₂O, which takes a finite time (minutes) to reach equilibrium.

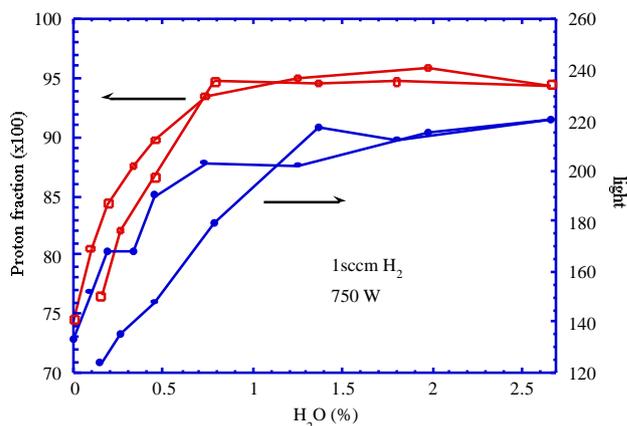


Figure 3: Proton fraction and light vs percent water in the source.

Figure 4 displays the proton fraction (H⁺/(H⁺ + H₂⁺ + H₃⁺)) and H emission from source with ~1% H₂O additive vs magnetic field. Light and proton fraction do not correlate well around the resonance condition. Indeed, maximum proton fraction is obtained at the resonance condition indicated by the arrow, where light is at a minimum. This probably occurs because efficient ionization of H atoms at resonance reduces their concentration and hence the intensity of H emission.

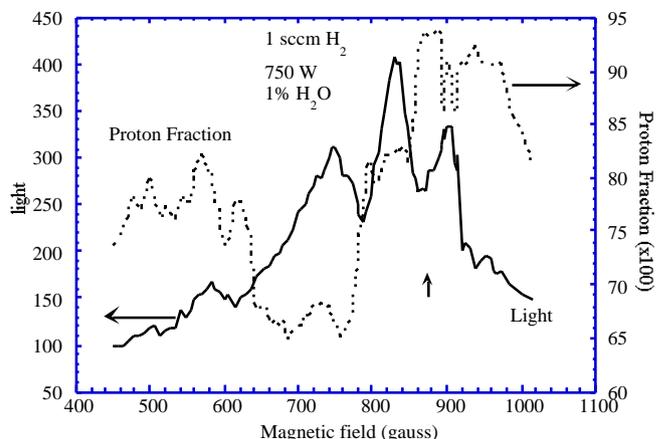


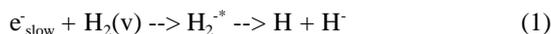
Figure 4: Light and proton fraction vs magnetic field.

b) Deuteron Production

Although not illustrated here, similar increases can be achieved in the deuteron fraction (D⁺/(D⁺ + D₂⁺ + D₃⁺)) by addition of H₂O or D₂O to D₂ plasmas, though this increase is less than in the case of H₂ as already high deuteron fraction can be obtained in pure D₂. Maximum deuteron fractions of about 0.93 are obtained. D₂O and H₂O are equally effective, although the use of H₂O results in minor impurities (H⁺, HD⁺) from ion-molecule reactions in the source.

c) H⁻ Production

Production of H⁻ ions in ion sources is known to occur by dissociative electron attachment of slow electrons to vibrationally excited hydrogen, H₂(v).



where H₂^{-*} is an intermediate temporary negative ion state. It is generally accepted that H₂(v) may be formed by the reaction

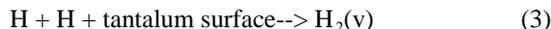


Attempts at direct extraction of negative ions from the standard magnetically-confined microwave-driven ion source described above by ourselves and others [8] have proved fruitless.

This probably arises from several causes. First, the mean energy of the electrons is generally low (~20 eV, Taylor and Wills [3]) and is not conducive to production of H₂(v) via equation (2). Second, the electrons in the source are magnetically confined along the axis between the extraction aperture and the microwave window. Thus, even if the weakly bound (0.75 eV) H⁻ species were formed in the source, it likely would be destroyed by collisional detachment by the 20 eV electrons.

However, Hall et. al. [7] have demonstrated alternative techniques for producing H₂(v), the precursor for H⁻. They find that H₂(v) can be generated in copious quantities by recombinative desorption of H atoms at surfaces coated with tantalum atoms. We have made use of this observation of Hall et. al. to achieve the first direct extraction of H⁻ ions from a magnetically confined microwave driven ion source.

In order to produce H₂(v), the precursor of H⁻, we have lined the inside of our source with a tantalum sheet to produce H₂(v), via the reaction



In order to reduce the flux of 20 eV electrons at the extraction aperture, the source solenoids were moved back from being centered on the source to being centered on the microwave window. This results in the magnetic field diverging and thus diverting electrons at the extraction aperture. Immediate extraction of H⁻ ions was achieved as measured by the QMS.

Electrons necessarily extracted along with the negative ions were removed from the beam prior to mass analysis by a weak (~30 Gauss) transverse magnetic field in the region of the accel-decel electrodes. About 100 mA of extracted electrons were collected on these electrodes.

Shown in Figure 5 is a mass spectrum of H⁻ extracted from the source. Based on the ratio of total H⁺ extracted to H⁻ sampled by the mass spectrometer, we estimate this first successful attempt to produce H⁻ resulted in a total H⁻ current

of 4-5 mA being extracted from the 5 mm diameter source aperture.

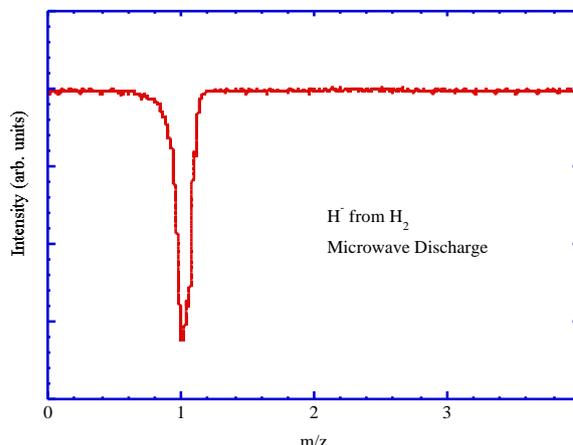


Figure 5: Mass spectrum of H⁻ extracted from the modified source.

This first data was obtained on August 20, 1996, and so optimal conditions for H⁻ production have not yet been achieved. Effects of additives have not yet been studied.

Acknowledgments

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